

Study Guide B

Answer Key

SECTION 1. ATOMS, IONS, AND MOLECULES

1. An element is a certain type of atom.
2. Sketch should resemble one of the illustrations in Figure 1.2 in Section 1. Nucleus should be in the center with protons and neutrons. Electrons should be labeled in a ring surrounding the nucleus.
3. A compound is composed of atoms of different elements.
4. an atom that has gained or lost one or more electrons
5. the number of protons no longer equals the number of electrons
6. Sketch should resemble the first illustration in Figure 1.3, with one atom losing an electron (becoming a positive ion) and the other accepting that electron (becoming a negative ion).
7. A bond that forms between two atoms that share a pair of electrons.
8. the number of electron pairs that it needs to share to fill its outer energy level
9. molecule
10. compound
11. element
12. ion
13. Ionic bonds form between oppositely-charged ions; covalent bonds form when atoms share a pair of electrons.

SECTION 2. PROPERTIES OF WATER

1. a molecule with a slightly positively charged region and a slightly negatively charged region
2. The oxygen nucleus pulls the electrons in the molecule more strongly than the hydrogen atoms. The hydrogen atoms have a slight positive charge and the oxygen atom has a slight negative charge.
3. an attraction between a slightly positive hydrogen and a slightly negative atom
4. between the slightly positive hydrogen atom of one water molecule and the slightly negative oxygen atom of another water molecule
5. A large amount of energy is needed to produce an increase in water temperature; water resists changes in temperature.
6. Water molecules "stick" to each other.
7. Water molecules "stick" to other substances.
8. A solvent is present in greater concentration and dissolves the solute.
9. ions and polar molecules
10. nonpolar molecules, such as fats and oils
11. **Effect on H^+ concentration:** acids increase H^+ concentration; bases decrease H^+ concentration
Effect on pH: acids lower pH; bases raise pH
12. Sketch should resemble Visual Vocab in Section 2 of the text. Solution: homogeneous mixture of substances; solvent: substance present in greater amount in which other substances dissolve; solute: substance that dissolves in a solvent.

SECTION 3. CARBON-BASED MOLECULES

1. Carbon atoms are the basis of the molecules that make up most living things.
2. Carbon atoms can form covalent bonds with up to four other atoms, including other carbon atoms.
3. Sketches should resemble Figure 3.1 in Section 3.
4. broken down as a source of chemical energy; part of cell structure
5. sugars, starches, cellulose
6. broken down as a source of chemical energy; part of cell structure
7. fats, oils, phospholipids
8. many functions, including movement, transport, chemical catalysts
9. enzymes, hemoglobin
10. store genetic information, build proteins

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Ian Pickup



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1. Understanding the eBook Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key
 - The Rise of Digital Reading Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key
 - Personalized Recommendations
 - Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key User Reviews and Ratings
 - Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key and Bestseller Lists
5. Accessing Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key Free and Paid eBooks
 - Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key Public Domain eBooks
 - Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key eBook Subscription Services
 - Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key Budget-Friendly Options

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 - ePub, PDF, MOBI, and More
 - Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key Compatibility with Devices
 - Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key
 - Highlighting and Note-Taking Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key
 - Interactive Elements Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key
8. Staying Engaged with Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key
9. Balancing eBooks and Physical Books Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key
 - Setting Reading Goals Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key
 - Fact-Checking eBook Content of Holt Mcdougal Chapter 6 Extra Skills Practice Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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