

Information Security Management Handbook

Sixth Edition

Edited by

Harold F. Tipton, CISSP · Micki Krause, CISSP

VOLUME 2



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motionless as that of an idol the narrator notes the town s pity for emily at this point in a discussion of her family s past the
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web sep 20 2014 i will illustrate this with some examples when an acid is neutralised by an alkali the reaction is exothermic eg 1 $\text{H}^+ \text{Cl}^- \text{aq} + \text{OH}^- \text{aq} \rightarrow \text{H}_2\text{O} \text{l}$ for which $\Delta H = 57 \text{ kJ mol}^{-1}$ eg 2 $\text{H}^+ \text{NO}_3^- \text{aq} + \text{OH}^- \text{aq} \rightarrow \text{H}_2\text{O} \text{l}$ for which $\Delta H = 57 \text{ kJ mol}^{-1}$

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web the enthalpy change of neutralisation is defined as the enthalpy change when solutions of an acid and alkali react together to produce one mole of water for example $\text{HCl aq} + \text{NaOH aq} \rightarrow \text{NaCl aq} + \text{H}_2\text{O aq}$ you can see that one mole of water is produced when one mole of HCl or one mole of NaOH reacts