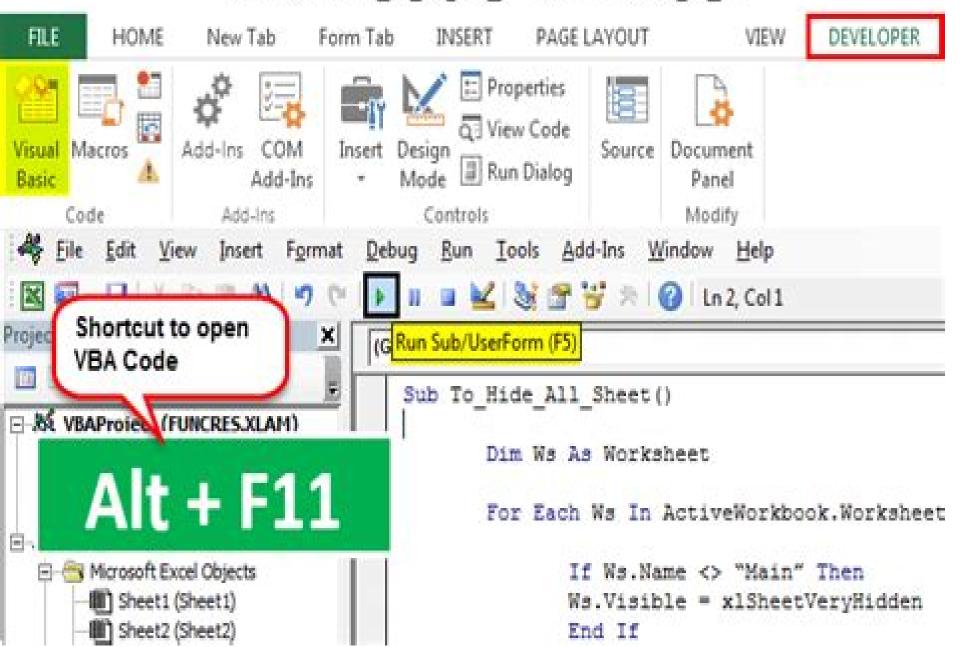
# **VBA Code in Excel**



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**Lingsheng Yao** 

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web nuclear reactions can be represented by balanced equations of nuclei in the azx form azx notation for atomic nuclei the top number a represents the nucleon number or the mass number nucleon number a total number of protons and neutrons in the nucleus the lower number z represents the proton or atomic number

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web to describe a nuclear reaction we use an equation that identifies the nuclides involved in the reaction their mass numbers and atomic numbers and the other particles involved in the reaction types of particles in nuclear reactions many entities can be involved in nuclear reactions

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web the carbon 14 nucleus is unstable it s radioactive it s going to undergo spontaneous decay it s going to try to get a better neutron to proton ratio so let s look at the nuclear equation which represents the spontaneous decay of carbon 14 so here is **21 2 nuclear equations chemistry 2e openstax** - Oct 09 2023

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web in alpha decay an alpha particle is ejected from an unstable nucleus so here s our unstable nucleus uranium 238 an alpha particle has the same composition as a helium nucleus we saw the helium nucleus in the previous video there are two protons in the helium nucleus and two neutrons

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web sep 12 2022 the half life of the nuclei and thus the material is t1 2 693 t 1 2 693 million years 35 a the decay constant is  $\lambda$  1 99 10 5s 1  $\lambda$  1 99 10 5 s 1 b since strontium 91 has an atomic mass of 90 90 g

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web nuclear equations represent the reactants and products in radioactive decay nuclear fission or nuclear fusion instead of chemical equations where it shows the different number of elements is conserved in a reaction in a nuclear reaction the atomic mass and proton number are conserved