

ACIDS, BASES AND SALTS

Definition of an acid An acid is a compound which when dissolved in water produces hydrogen ions as the only positively charged ions	What causes the acidic properties of acids? The hydrogen ions (H^+) cause the acidic properties and these are formed in the presence of water.
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Another term to refer to an acid An acid is called a <u>proton donor</u>	Why an acid is also called a proton donor? It's because an acid provides protons or hydrogen ions (H^+) to other substances during the reaction.
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Substance to which an acid provides protons BASE	A base is then called PROTON ACCEPTOR	Why the base is called a proton acceptor? Because the base accepts hydrogen ions from acids	Equation for the reaction between acid and base $H^+_{(aq)} + OH^-_{(aq)} \longrightarrow H_2O_{(l)}$
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Common laboratory acids		These three common laboratory acids are also called	Why these acids are also called mineral acids?
Hydrochloric acid	HCl	MINERAL ACIDS	They are derived from mineral salts i.e. chlorides for HCl, sulphates for H_2SO_4 and nitrates for HNO_3
Sulphuric acid	H_2SO_4		
Nitric acid	HNO_3		

Other mineral acids known		Mineral salts from which the acid is derived	Organic acids known	Naturally occurring acids
Sulphurous acid	H_2SO_3	Derived from sulphites	Ethanoic acid (CH_3COOH)	CITRIC ACID from lemons
Carbonic acid	H_2CO_3	Derived from carbonates		TARTARIC ACID from grapes
Phosphoric acid	H_3PO_4	Derived from phosphates	Methanoic acid ($HCOOH$)	ACETIC ACID from vinegar
Nitrous acid	HNO_2	Derived from nitrites		LACTIC ACID from sour milk
				Hydrochloric acid from digestive juices

Whenever an acid is dissolved in water, it produces HYDROGEN IONS	Term given to the number of hydrogen ions produced by one molecule of an acid BASICITY OF AN ACID	Definition of basicity of an acid BASICITY of an acid is the <u>number of hydrogen ions</u> produced by <u>one molecule</u> of an acid <u>in aqueous solution</u> .
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Guide Acids Bases And Salts

Annelies Wilder-Smith



Guide Acids Bases And Salts:

This book delves into Guide Acids Bases And Salts. Guide Acids Bases And Salts is an essential topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Guide Acids Bases And Salts, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Guide Acids Bases And Salts
- Chapter 2: Essential Elements of Guide Acids Bases And Salts
- Chapter 3: Guide Acids Bases And Salts in Everyday Life
- Chapter 4: Guide Acids Bases And Salts in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Guide Acids Bases And Salts. This chapter will explore what Guide Acids Bases And Salts is, why Guide Acids Bases And Salts is vital, and how to effectively learn about Guide Acids Bases And Salts.
3. In chapter 2, this book will delve into the foundational concepts of Guide Acids Bases And Salts. The second chapter will elucidate the essential principles that must be understood to grasp Guide Acids Bases And Salts in its entirety.
4. In chapter 3, this book will examine the practical applications of Guide Acids Bases And Salts in daily life. This chapter will showcase real-world examples of how Guide Acids Bases And Salts can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Guide Acids Bases And Salts in specific contexts. This chapter will explore how Guide Acids Bases And Salts is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Guide Acids Bases And Salts. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Guide Acids Bases And Salts.

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