

CHAPTER 6 REVIEW*Chemical Bonding***MIXED REVIEW**

SHORT ANSWER Answer the following questions in the space provided.

1. Name the type of energy that is a measure of strength for each of the following types of bonds:

_____ lattice energy _____	a. ionic bond
_____ bond energy _____	b. covalent bond
_____ enthalpy of vaporization _____	c. metallic bond

2. Use the electronegativity values shown in **Figure 20**, on page 161 of the text, to determine whether each of the following bonds is nonpolar covalent, polar covalent, or ionic.

_____ ionic _____	a. H—F	_____ nonpolar covalent _____	d. H—H
_____ ionic _____	b. Na—Cl	_____ polar covalent _____	e. H—C
_____ polar covalent _____	c. H—O	_____ polar covalent _____	f. H—N

3. How is a hydrogen bond different from an ionic or covalent bond?

A hydrogen bond is a dipole-dipole attraction between a partially positive hydrogen
atom and the unshared electron pair of a strongly electronegative atom such as O,
N, or F. Unlike ionic or covalent bonds, in which electrons are given up or shared,
the hydrogen bond is a weaker attraction. Hydrogen bonds are generally
intermolecular, while ionic and covalent bonds occur between ions or atoms
respectively.

4. H₂S and H₂O have similar structures and their central atoms belong to the same group. Yet H₂S is a gas at room temperature and H₂O is a liquid. Use bonding principles to explain why this is.

Oxygen has higher electronegativity than sulfur, which creates a highly polar bond.
Increased polarity in H₂O bonds means a stronger intermolecular attraction, making
water a liquid at room temperature. Hydrogen bonding exists between water
molecules, but not between hydrogen sulfide molecules.

Guided And Study Workbook Answers Covalent Bonding

Judd E. Hollander



Guided And Study Workbook Answers Covalent Bonding:

The book delves into Guided And Study Workbook Answers Covalent Bonding. Guided And Study Workbook Answers Covalent Bonding is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Guided And Study Workbook Answers Covalent Bonding, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Guided And Study Workbook Answers Covalent Bonding
 - Chapter 2: Essential Elements of Guided And Study Workbook Answers Covalent Bonding
 - Chapter 3: Guided And Study Workbook Answers Covalent Bonding in Everyday Life
 - Chapter 4: Guided And Study Workbook Answers Covalent Bonding in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, the author will provide an overview of Guided And Study Workbook Answers Covalent Bonding. This chapter will explore what Guided And Study Workbook Answers Covalent Bonding is, why Guided And Study Workbook Answers Covalent Bonding is vital, and how to effectively learn about Guided And Study Workbook Answers Covalent Bonding.
 3. In chapter 2, the author will delve into the foundational concepts of Guided And Study Workbook Answers Covalent Bonding. The second chapter will elucidate the essential principles that must be understood to grasp Guided And Study Workbook Answers Covalent Bonding in its entirety.
 4. In chapter 3, this book will examine the practical applications of Guided And Study Workbook Answers Covalent Bonding in daily life. The third chapter will showcase real-world examples of how Guided And Study Workbook Answers Covalent Bonding can be effectively utilized in everyday scenarios.
 5. In chapter 4, the author will scrutinize the relevance of Guided And Study Workbook Answers Covalent Bonding in specific contexts. This chapter will explore how Guided And Study Workbook Answers Covalent Bonding is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, the author will draw a conclusion about Guided And Study Workbook Answers Covalent Bonding. The final chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Guided And Study Workbook Answers Covalent Bonding.

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