

# Stoichiometry Guided Practice:

Name \_\_\_\_\_

Period \_\_\_\_ Date \_\_\_\_/\_\_\_\_/\_\_\_\_

## Molar to Molar Stoichiometry:



- What is the mole ratio of  $\text{CaSO}_3$  to water?
- How many moles of  $\text{CaSO}_3$  will be produced if 4 moles of water are also produced?
- What is the mole ratio of  $\text{Ca(OH)}_2$  to water?
- How many moles of water are produced if 4 moles of  $\text{Ca(OH)}_2$  are used in the reaction?



- What is the mole ratio of water to  $\text{CaCO}_3$ ?
- How many moles of  $\text{H}_2\text{O}$  will be produced if 0.2000 moles of  $\text{CaCO}_3$  are produced?
- What is the mole ratio of  $\text{CaCO}_3$  to  $\text{NH}_3$ ?
- How many moles of  $\text{NH}_3$  will be produced if 0.5 moles of  $\text{CaCO}_3$  are used in the reaction?



- What is the mole ratio of  $\text{Cl}_2$  to  $\text{PCl}_3$ ?
- How many moles of  $\text{Cl}_2$  will react with 0.4624 moles of  $\text{PCl}_5$ ?

# Guided Study Work Answers Stoichiometry

**Wade A. Freeman**



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**Robinson Chemistry Study Guide** Robinson, 1992

*Student Study Guide Fundamentals of Chemistry by David E. Goldberg* Margaret G. Kimble, 1994

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