



Oil Cooler, Radiator, Universal Joint and Removing Hydraulic Pump

1. Remove the bonnet cover (1) and ROPS connecting plate.
2. Remove the oil cooler (2).
3. Remove the radiator (3).
4. Remove the ROPS connecting plate, under (4).
5. Remove the hydraulic pump (5).

(When reassembling)

Tightening torque	ROPS connecting plate mounting screw (M8)	24 to 27 N·m 2.4 to 2.8 kgf·m 18 to 20 lbf·ft
	ROPS connecting plate, under mounting screw (M10)	48 to 55 N·m 4.9 to 5.7 kgf·m 35 to 41 lbf·ft
	Hydraulic pump mounting screw	40 to 44 N·m 4.0 to 4.5 kgf·m 29 to 32 lbf·ft

- (1) Bonnet Cover
(2) Oil Cooler
(3) Radiator

- (4) ROPS Connecting Plate, Under
(5) Hydraulic Pump

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Kubota 326 Parts Manual

AW Chickering



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density of cs 7 85 g cm³ the volume of water required for hydro testing is estimated as $\pi \times \text{pipe id in meter}^2 \times \text{length of pipe}$
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web determine the required pipe size of each section and outlet of the piping system shown in figure a 6 1 with a designated pressure drop of 0 5 inch w c 125 pa using the longest length method the gas to be used has 0 60 specific gravity and a heating value of 1 000 btu ft 3 37 5 mj m 3 solution maximum gas demand for outlet a the length of pipe

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web procedure 1 the relationship between pressure and volume set the plunger of the syringe to 1 10 the capacity of the syringe the 2 0 ml mark if using a 20 ml syringe the volume is read from the edge of the bottom rib of the plunger attach the syringe to the connector at the end of sensor

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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