

Step 2: Calculate the ages of the metamorphic and igneous layers based on the following information.

- A mineral sample from rock unit B has 150,000 atoms of uranium-235 and 50,000 atoms of lead-207.
- A mineral sample from rock unit C has 180,000 atoms of uranium-235 and 20,000 atoms of lead-207.
- A mineral sample from rock unit G has 100,000 atoms of potassium-40 and 100,000 atoms of argon-40.

Use **Figure 9.8**, if necessary, to help you determine calculate how many half-lives have passed since rock units B, C, and G were each formed, based on the remaining amount of a parent isotope compared to the original amount of parent isotope.

Recall that the original amount of parent isotope can be calculated by adding together the amount of parent isotope plus the amount of daughter isotope, making the assumption that neither has escaped the sample. Also note that if less time has passed than a full half-life, you must calculate what portion of a half-life has passed to make the calculations.

Once you know how many half-lives have past since the sample crystallized, use the half-life table in **Figure 9.9** to calculate an absolute date in mya (million years ago) for each rock unit.

11. The igneous rock in rock unit B was formed _____.
12. The igneous rock in rock unit C was formed _____.
13. The metamorphic rock in rock unit G was formed _____.

Step 3: Index fossils have been found in the sedimentary layers! Use **Figure 9.11** to help you to identify the periods in which layers A, D, E, and F were formed as well as their age in mya based on the discovery of the following fossils:

14. The following fossils were found in Layer A:



Figure 9.13

Layer A was formed _____ mya during the _____ Period.

15. The following fossils were found in Layer D:



Figure 9.14

Layer D was formed _____ mya during the _____ Period.

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