Experiment 14

Data and Calculations: Heat Effects and Calcrimetry

A. Specific Heat

Trial 1		Trial 2
45,393 8		45,406 =
22, 525 8	-	22(139p 8
	-	S. 68 8
LIT.798 8		48.2918
37,175 #		31,687 =
228 68 =		_22.8(8_g
21,7 %		2150 0
99.8 ~		_(0013_°c
2513 °c		2415 °C
°c		°c
~c		°c
		3/g°C
	264.	
	45,393 = 22,525 = 3998 = 39,117 = 39,117 = 22,8 18 = 21,17 = c 99,18 = c 90,18 = c 90,18 = c	45,393 * - 22,1725 * - 3,167 * - 41,173 * - 41,173 * - 41,173 * 41,1748 * - 22,8 68 * - 21,12 * c - 25,13 * c - *

B. Heat of Solution

Mass of calorimeter plus water

Mass of beaker

29.309 =

(continued on following page)

Lab Manual Chemistry Heat Effects And Calorimetry

John Francis Woodhull, May Belle Van Arsdale **Lab Manual Chemistry Heat Effects And Calorimetry:**

Whispering the Techniques of Language: An Psychological Quest through **Lab Manual Chemistry Heat Effects And Calorimetry**

In a digitally-driven world where monitors reign great and quick interaction drowns out the subtleties of language, the profound techniques and mental nuances hidden within phrases frequently get unheard. Yet, located within the pages of **Lab Manual Chemistry Heat Effects And Calorimetry** a captivating literary prize blinking with natural thoughts, lies a fantastic quest waiting to be undertaken. Composed by an experienced wordsmith, this charming opus encourages viewers on an introspective trip, gently unraveling the veiled truths and profound impact resonating within the very cloth of each and every word. Within the emotional depths of this emotional evaluation, we will embark upon a heartfelt exploration of the book is primary themes, dissect their charming publishing model, and succumb to the effective resonance it evokes serious within the recesses of readers hearts.

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