11.3. For the excitation transfer process indicated in Fig. 11.1b, write rate equations like (11.3.3) for the populations N_A, N_B, and N_{B*}.

We can write the following rate equation for the number density N_{A^*} of excited atoms A^* (or molecules) of species A due to the process (11.3.1):

$$\frac{dN_{A^*}}{dt} = -R(N_{A^*}N_B - e^{-\Delta E/k_BT}N_AN_{B^*}). \tag{11.3.3}$$

It is conventional to designate an excited atom or molecule by an asterisk. The exothermic process indicated in Fig. 11.1a is written out as a reaction as follows:

$$A^* + B \longrightarrow A + B^* + \Delta E$$
. (11.3.1)

Likewise the endothermic process of Fig. 11.1b is written symbolically as

$$A^* + B \longrightarrow A + B^* - \Delta E$$
. (11.3.2)



Figure 11.1 (a) Exothermic and (b) endothermic excitation from atom A to atom B.

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