

#### EXPERIMENT NO.1 EXPERIMENT NAME: Redox Titration

Objective: - Find out the percentage of iron in a given iron alloy solution, when X grams of affiny stimustyand in 1 frage

Apparatus/Reagent Required :- Nurvette, Properte, Control Starte, Parantal Stall NJ 90 KM/mO., satisface., dill H,5O., Alloy solution.

#### Theory: -

To determine the amount of iron in along the alloy sample is dissolved in dil. H.SO., the iron present is converted to ferrous sulfate with the liberation of hydrogen gas.

The Fe<sup>3-1</sup> present in the solution can be determined by titration with a standard KMnD, solution. Potassium permanganate Oxidizes femous ion into femic ion in acid medium in cold and is mail reduced. to colourless manganous ion.

(West-States)

The appearance of a faint but permanent gink colour marks the end point.

#### Chemical Equations-

28MmO,+8H,5O,+10FeSO,-9 K,5O,+2MmSO,+5Fe;(SO,),+8H/O

#### Procedure:

Fell buserine with KMoOL solution.

# **Lorven Publications Engg Chemistry Lab Manuals**

Jesse Yeh

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