



542011 Bleeding the fuel system (see Figure 7) 8140.43C - 8140.67 engines only

Loosen bleed screw (12), operate priming pump lever (13) until complete bleeding of all the air in the system, tighten bleed screw.



If the engine stops for lack of fuel, resulting in air being introduced into the system (and if the bleeding process described above is not sufficient), it is necessary to loosen the unions of at least two injectors, rotate the engine and start it.

When the air has been bled off, close the unions again.



501430 Power steering system air bleed

Check the level of oil in the tank and top it up if necessary. Lift the vehicle at the front, start up the engine and let it idle for some time.

Check there is no oil leakage from the hydraulic circuit and check the level in the tank.

Slowly turn the flywheel in both directions of steering so that the air in the hydraulic system comes out.

Check the level of oil in the tank again and top up if necessary.

507130 OXICAT - OXYDIZER CATALYST (Vehicles with 8140.43C - 8140.63 engines)

Description

The oxidizing catalyst (1) is a post-treatment device for exhaust gases. It is used to oxidize CO, HC transforming them into carbon dioxide (CO_2) and steam (H_2O).

This catalyst also treats saturated and aromatic hydrocarbons forming part of the particulate, such as the carbon in the form of soot, metals, water and sulphur compounds that are exhausted.

The cylindrical-shaped catalyst consists of a ceramic honeycomb monolith whose cells are platinum-impregnated, a catalyzing substance of oxidation reactions. Its total volumetric capacity is equal to 600 cm^3 with a density of 400 cells per square inch.

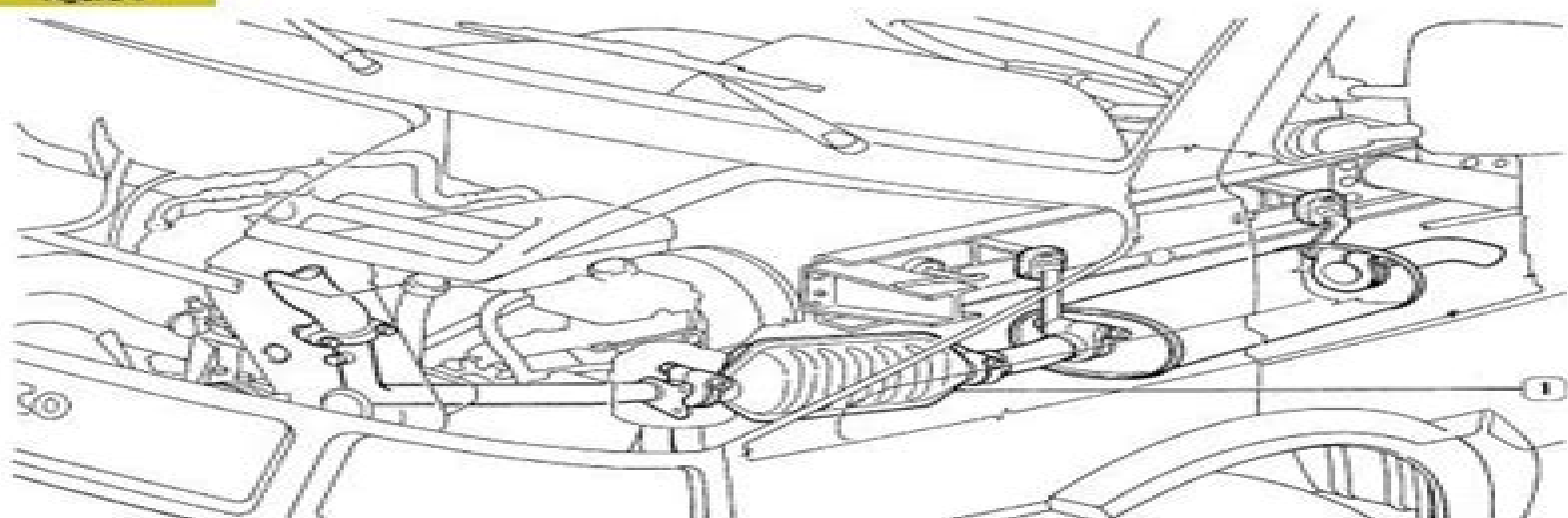
The exhaust gases passing through the cells heat the catalyst converting the pollutants into inert compounds.

The chemical oxidation reaction of CO, HC and of the particulate is efficient with temperatures ranging between 300°C - 350°C .

With temperatures higher than 350°C it starts oxidizing the sulphur contained in the diesel oil, generating sulphur dioxide (SO_2) and sulphur trioxide (SO_3), which are the cause of acid rain.

Correct catalyst sizing enables limiting the temperature, obtaining the greatest percentage of conversion of polluting emissions, while reducing the freeing of sulphur compounds.

Figure 9



401-01

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L Reisser



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