

Name _____ Date _____ Class _____

CHAPTER 9

STUDY GUIDE FOR CONTENT MASTERY

Covalent Bonding

Section 11 The Covalent Bond

In your textbook, read about the nature of covalent bonds.

Use each of the terms below just once to complete the passage.

covalent bond	molecule	sigma bond	exothermic	π bond
---------------	----------	------------	------------	------------

When sharing electron pairs, the attachment between atoms that results is called a(n) **covalent bond**. When such an attachment is formed, bond dissociation energy is released, and the process is **exothermic**. When two or more atoms bond by means of electron sharing, the resulting particle is called a(n) **molecule**. If the electrons shared are contained between the two atoms, the attachment is called a(n) **sigma bond**. If the sharing involves the overlap of parallel orbitals, the attachment is called a(n) **π bond**.

In your textbook, read about single and multiple bonds and bond strengths.

Circle the letter of the choice that best completes the statement or answers the question.

- Is water from its elements such as hydrogen, oxygen, and oxygen normally **exothermic**?
 - an single atom
 - an molecules containing three atoms
 - an molecules containing two atoms
 - an molecules containing four atoms
- How many electrons are shared in a double covalent bond?
 - none
 - two
 - four
 - eight
- Bond length is the distance between
 - two molecules of the same substance
 - the electrons in two attached atoms
 - the nuclei of two attached atoms
 - the orbitals of two attached atoms
- Which of the following relationships relating to bond length is generally correct?
 - the shorter the bond, the stronger the bond
 - the shorter the bond, the weaker the bond
 - the shorter the bond, the fewer the electrons in it
 - the shorter the bond, the lower the bond dissociation energy

Study Guide for Content Mastery

Chemistry: Matter and Change • Chapter 9

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Name _____ Date _____ Class _____

CHAPTER 9 STUDY GUIDE FOR CONTENT MASTERY

Section 12 Naming Molecules

In your textbook, read about how binary compounds and acids are named from their formulas.

For each statement below, write *true* or *false*.

- false** _____ 1. Binary molecular compounds are generally composed of a metal and a nonmetal.
- false** _____ 2. The second element in the formula of a binary compound is named using the suffix *-ite*.
- false** _____ 3. The prefix *mono-* indicates three atoms.
- true** _____ 4. The prefix *hepta-* indicates six atoms.
- true** _____ 5. In naming the first element in a formula, the prefix *mono-* is not used.
- true** _____ 6. For binary acids, the hydrogen part of the compound is named using the prefix *hydro-*.
- false** _____ 7. An oxyacid contains only two elements.
- false** _____ 8. If the name of the cation of an oxyacid ends in *-ite*, the acid name contains the suffix *-ate*.

In your textbook, read about naming molecular compounds and oxyacids.

For each item in Column A, write the letter of the matching item in Column B.

Column A

- c.** _____ 9. CO
- f.** _____ 10. CO₂
- g.** _____ 11. H₂CO₃
- e.** _____ 12. N₂
- b.** _____ 13. N₂O₄
- d.** _____ 14. HNO₃
- f.** _____ 15. HNO₂
- a.** _____ 16. HBr
- h.** _____ 17. HBrO₃

Column B

- a.** hydrobromic acid
- b.** nitrogen tetroxide
- c.** carbon monoxide
- d.** nitrous acid
- e.** carbonic acid
- f.** nitric acid
- g.** carbonic acid
- h.** bromic acid
- i.** carbon dioxide

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Key Minerals Study Guide For Content Mastery

Julia Schneider



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Electrical Measuring Instruments and Measurements S.C. Bhargava, 2012-12-27 This book written for the benefit of engineering students and practicing engineers alike is the culmination of the author's four decades of experience related to the subject of electrical measurements comprising nearly 30 years of experimental research and more than 15 years of teaching at several engineering institutions The unique feature of this book apart from covering the syllabi of various universities is the style of presentation of all important aspects and features of electrical measurements with neatly and clearly drawn figures diagrams and colour and b w photos that illustrate details of instruments among other things making the text easy to follow and comprehend Enhancing the chapters are interspersed explanatory comments and where necessary footnotes to help better understanding of the chapter contents Also each chapter begins with a recall to link the subject matter with the related science or phenomenon and fundamental background The first few chapters of the book comprise Units Dimensions and Standards Electricity Magnetism and Electromagnetism and Network Analysis These topics form the basics of electrical measurements and provide a better understanding of the main topics discussed in later chapters The last two chapters represent valuable assets of the book and relate to a Magnetic Measurements describing many unique features not easily available elsewhere a good study of which is essential for the design and development of most electric equipment from motors to transformers and alternators and b Measurement of Non electrical Quantities dealing extensively with the measuring techniques of a number of variables that constitute an important requirement of engineering measurement practices The book is supplemented by ten appendices covering various aspects dealing with the art and science of electrical measurement and of relevance to some of the topics in main chapters Other useful features of the book include an elaborate chapter by chapter list of symbols worked examples exercises and quiz questions at the end of each chapter and extensive authors and subject index This book will be of interest to all students taking courses in electrical measurements as a part of a B Tech in electrical engineering Professionals in the field of electrical engineering will also find the book of use

Mastering Artisan Cheesemaking Gianaclis Caldwell, 2012-09-19 Every serious home scale artisan cheesemaker even

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