

## UNIT (6) ACIDS AND BASES

### 6.1 | Arrhenius Definition of Acids and Bases

Definitions for acids and bases were proposed by the Swedish chemist *Savante Arrhenius* in 1884.

**Acids** were defined as compounds that contain hydrogen and can dissolve in water to release hydrogen ions into solution. For example, HCl dissolves in water as follows:



**Bases** are defined as substances that dissolve in water to release hydroxide ions ( $\text{OH}^-$ ) into solution. For example, a typical base according to the Arrhenius definition is sodium hydroxide, NaOH:



The process that converts a molecule such as HCl into ions is called ionization.

**Ionization** is the production of ions from molecular compounds. No ions were initially present.

The process of converting a solid ionic compound, such as NaOH, into ions in solution is called dissociation.

**Dissociation** is the production of ions from ionic compounds. Ions were initially present, water just breaks apart ions.

**Some common acidic and basic substances:**

Acidic	Basic
citrus fruits	soap
vinegar	detergents
aspirin	milk of magnesia

**Some characteristic properties associated with acids and bases:**

Acids	Bases
sour taste	bitter taste
change litmus paper to red	change litmus paper to blue
react with hydroxide bases to produce salt	slippery, soapy feeling

# Harcourtschoolsupply Answers Unit 1

**Puja Mehta**



## **Harcourtschoolsupply Answers Unit 1:**

## **Harcourtschoolsupply Answers Unit 1** Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the energy of words has become more evident than ever. They have the ability to inspire, provoke, and ignite change. Such is the essence of the book **Harcourtschoolsupply Answers Unit 1**, a literary masterpiece that delves deep to the significance of words and their affect our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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