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Logo

FULL, BLACK



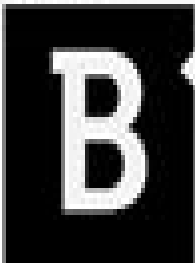
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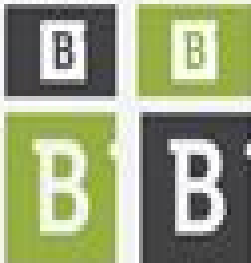
FULL, COLOR



ICON, BLACK



ICON, VARIATIONS



Font

PRIMA NOVA

ABCDEFGHIJKLMNOPQRSTUVWXYZ
abcdefghijklmnopqrstuvwxyz

ALL WEIGHTS

Light, Light Italic, Regular, Italic, Semibold,
Semibold Italic, Bold, Bold Italic, Extrabold,
Extrabold Italic, Black

Colors*

A, BRAFTON GREEN

VALUES: C 000
M 100 Y 000
K 100 P 100
R 000 G 000
#000000 #000000

B, BLUE

VALUES: C 000
M 100 Y 000
K 100 P 100
R 000 G 000
#000000 #000000

C, BLACK

VALUES: C 100
M 100 Y 100
K 100 P 100
R 000 G 000
#000000 #000000

*All colors may have variations, which may
contribute to color theory combinations.



Graphic Style Guide Example

N Colangelo



Graphic Style Guide Example:

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web where the salt is the conjugate base of the acid the four parts of the titration curve are described below and you should look to the appropriate text section to see how they are treated pure acid 0 mL of base is added section 17.3.2.1

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web solution a titrant volume 0 mL the solution ph is due to the acid ionization of HCl because this is a strong acid the ionization is complete and the hydronium ion molarity is 0.100 M the ph of the solution is then $ph = -\log 0.100 = 1.000$ ph $\log 0.100 = 1.000$ b titrant volume 12.50 mL

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web at the equivalence point in an acid base titration moles of base = moles of acid and the solution only contains salt and

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web ab acid base titration curves purpose the purpose of this lab is to determine the shape of acid base titration curves theoretically and experimentally and to use an acid base titration to measure the effectiveness of antacid tablets summary a titration curve is the plot of the pH value of a solution vs the amount of titrant added

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Volume of NaOH added (mL)	pH of the solution
4	6.4
6	4.6
5	9.5
9	5.9

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