

Gregory F. Reeves

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SUMMARY

Thin Film Vacuum Coating Technician with extensive experience with e-beam vacuum coating chambers. Inspect all coating runs with Metrology tools (spectrophotometer & coating thickness). Prepare substrates for coating with etching chemicals. Use of glass cutting tools to cut substrates for coating runs. Operated engine lathes & Bridgeport mills as conventional Machinist, machining various metals. Knowledgeable in the use of measuring, hand & power tools. Received forklift training with certificate of completion. Use of bead-blasting equipment. Knowledge in Microsoft Office software.

PROFESSIONAL EXPERIENCE

ZsCoat Inc. - Torrance, CA

2014 - present

Thin Film Vacuum Coating Technician

- Operate E-beam ion-assist vacuum coating chamber, producing thin film optical coatings for mirrors & reflectors.
- Perform general maintenance and repair on vacuum coating chamber.
- Inspect coating runs using metrology tools (spectrophotometer).
- Meticulous record keeping.

Coursen Coating Labs - Newport Beach, CA

2010 - 2013

Thin Film Technician

- Operated E-beam vacuum coating chambers, producing thin film optical coatings (dichroic art glass, UV, HEHM, HECM, B-BAR, and AR).
- Created coating run set-ups and provided support to other techs.
- Processed "clean glass" for coating.
- Performed general maintenance and repairs on vacuum coating chambers.
- Inspected coating runs, using metrology tools (spectrophotometer).
- Maintained cleanliness of tooling & vacuum chamber parts using bead blaster equipment.
- Performed shipping and receiving of substrates & coated parts.

Coatings by Sandberg - Orange, CA

2001 - 2010

Assistant Production Supervisor

- Supervised, developed, coached, and evaluated vacuum coating technicians.
- Operated E-beam vacuum coating chambers, coating dichroic art glass.
- Performed general maintenance and repairs for E-beam vacuum coating chambers.
- Inspected production runs, using metrology tools (spectrophotometer) & data entry.
- Developed technique to reduce breakage of coated glass.
- Created parts and tooling, using machine shop equipment.
- Maintained cleanliness of tooling & vacuum chamber parts using bead blaster equipment.
- Processed "clean glass" for coating.
- Scheduled glass to be coated for both Glass Cutting and Production Departments.
- Purchased raw glass & glass cutting supplies for Glass Cutting Department.
- Transported glass between delivery trucks and warehouse, using forklift.

EDUCATION

Air Conditioning and Refrigeration Certificate, Long Beach City College, Long Beach, CA

Microsoft Office Suite Certificate, Garden Grove Adult Education, Garden Grove, CA

Greggs Manual Resume

Louis A. Leslie



Greggs Manual Resume:

Unveiling the Magic of Words: A Overview of "**Greggs Manual Resume**"

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web with a strong acid and strong base titration strong acids and strong bases refer to species that completely dissociate to form ions in solution by contrast weak acids and bases ionize only partially and the ionization reaction is reversible

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the

number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass
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web goals understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web titration experiment in the neutralization of hydrochloric acid by sodium hydroxide the mole ratio of acid to base is 1 1
 $\text{HCl(aq)} + \text{NaOH(aq)} \rightarrow \text{NaCl(aq)} + \text{H}_2\text{O(l)}$
 one mole of HCl would be fully neutralized by one mole of NaOH

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak pK_a of an unknown acid or pK_b of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being NaOH reaction with a 2nd substance $\text{H}_2\text{C}_2\text{O}_4 + 2\text{NaOH(aq)} \rightarrow \text{Na}_2\text{C}_2\text{O}_4 + 2\text{H}_2\text{O}$ to differentiate an acid from a base in a solution a pH scale is

used the ph scale ranges from 0 to 14 a ph of 7 is neutral in an acid base reaction

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web this point of titration is referred to as the equivalence point considering the 1 1 stoichiometry of this acid base reaction naoh aq c6h4 cooh cook aq c6h4 coona cook aq h2o l the point of equivalence is the point of titration when the number of moles of naoh na added is equal to the number of moles of khp nb in the

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