

12

STOICHIOMETRY

SECTION 12.1 THE ARITHMETIC OF EQUATIONS (pages 353–358)

This section explains how to calculate the amount of reactants required or product formed in a nonchemical process. It teaches you how to interpret chemical equations in terms of interacting moles, representative particles, masses, and gas volume at STP.

► Using Everyday Equations (pages 353–355)

1. How can you determine the quantities of reactants and products in a chemical reaction?

You can use the balanced equation.

2. Quantity usually means the amount of a substance expressed in grams or moles.

3. A bookcase is to be built from 3 shelves (Sh), 2 side boards (Sb), 1 top (T), 1 base (B), and 4 legs (L). Write a “balanced equation” for the construction of this bookcase.

$3\text{Sh} + 2\text{Sb} + \text{T} + \text{B} + 4\text{L} = \text{Sh}_3\text{Sb}_2\text{TBL}_4$

► Using Balanced Chemical Equations (page 354)

4. Is the following sentence true or false? Stoichiometry is the calculation of quantities in chemical reactions. true

5. Calculations using balanced equations are called stoichiometric calculations.

► Interpreting Chemical Equations (pages 356–357)

6. From what elements is ammonia produced? How is it used?

Ammonia molecules are composed of nitrogen and hydrogen; it is used as a fertilizer.

7. Circle the letter of the term that tells what kind of information you CANNOT get from a chemical equation.

a. moles

d. volume

b. mass

e. number of particles

☒ c. size of particles

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The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

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Study Guide for Pharmacology Joyce LeFever Kee, Evelyn R. Hayes, Linda E. McCuistion, 2014-01-30 This study guide contains a variety of exercises specifically designed to reinforce content presented in the textbook Chapter worksheets feature more than 1800 study questions 160 drug calculation and dosage problems and case based critical thinking exercises

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Math Problems and Solutions Guide David Scheinker, 2009

IIT JEE Mathematics Study Notes: Guide for Exam Preparation, 2024-04-01 A *Guide to Evidence-Based Group Work* Mark J. Macgowan, 2008-02-12 This is the first textbook that illustrates step by step how to practice evidence based group work As group workers are increasingly being held accountable to evaluate monitor and improve their practice there are scant resources available that apply specifically to their practice General books on evidence based practice lack the rich material on group work organized for the first time in this one volume Designed specifically as a supplement for undergraduate and graduate group work courses the text is organized around the process and philosophy of evidence based practice formulating appropriate practice questions searching for evidence critically reviewing available evidence applying the evidence with judgment skill and concern for the group members and evaluating the outcomes achieved The evidence based group work framework teaches students how to skillfully apply best practices through practical learning activities case examples research designs step by step guidelines and checklists assessment tips and detailed instructions for formulating questions and evaluating answers It will be equally valuable as a handy reference for professionals eager to hone their practice skills

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