

MCQ on Chemical Bonding

1. Multiple covalent bond containing compound is - **P₄**
2. The octet rule is observed in - **CO₂**
3. The van der Waal's force occurs in - **Liquid argon**
4. Which among the following does not have the hydrogen bond? - **Liquid HCl**
5. In diamond, the bond present is - **Covalent**
6. The bond angle of sp^2 -hybrid orbital is - **120°**
7. The types of bond present in CuSO₄. 5H₂O are only - **Electrovalent and covalent**
8. Molecular shape of PF₅, is - **Trigonal bipyramidal**
9. Ionic bond can easily be formed between - **Na and Cl**
10. Good electron pair donor is - **N(CH₃)₃**
11. Write the state of hybridization in PCl₅ - **sp^3d**
12. Which of the following has smallest bond angle? - **H₂O**
13. Sulphate ion is - **Tetrahedral**
14. The bond angle in water is - **104.5°**
15. Which of the following molecule is planar? - **C₂H₄**

Chemistry Objective Questions

Inorganic Chemistry Mcqs With Answers

Mark Steyvers



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