

Periodic Law

OBJECTIVES

- To observe and investigate the periodic trends for reactivity, density, pH and solubility within groups 1, 2 and 14 in the periodic table of the elements.

INTRODUCTION

When the elements are arranged in order of increasing atomic number, there is a periodic occurrence of properties that leads to the grouping of elements on the periodic table. Elements in vertical columns form groups or families that are characterized by marked similarities in physical and chemical properties. These similarities are due to the fact that all of the elements within a group have the same number of valence electrons. Within a group, periodic trends in certain properties can be observed. The properties of density and solubility, for example, are known to vary regularly within groups of elements.

PROCEDURE

PART A

The teacher will demonstrate the reactivity of three elements (Lithium, Sodium and Potassium) in group 1, by adding a small piece of the metal to water. Observe data sheet and record observations and rank the three elements in order of reactivity.

Element	Reactivity	Observations
Li		
Na		
K		

PART B

- Pour about 5 mL of distilled water into a clean dry test tube.
- FRONT LAB TABLE:** Remove the pH probe with caution from the water bottle and insert the pH probe into the distilled water. Note the pH on the "Lab Quest". Remove the pH probe and store with the water bottle.
- FRONT LAB TABLE:** Add a few drops of phenolphthalein solution to the reaction tube. Phenolphthalein solution turns pink in the presence of a base.

Instructors Manual For Periodic Law Experiment

Edward Bradford Titchener



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Quantitative experiments: pt. 1. Students' manual. pt. 2. Instructor's manual Edward Bradford Titchener, 1905

Experimental Psychology: Quantitative experiments: pt. 1. Students' manual. pt. 2. Instructor's manual Edward Bradford Titchener, 1905 The general purpose with which this book has been written is sufficiently indicated by its title I have selected a number of the classical experiments of Experimental Psychology and have tried to present them in such a way that their performance shall have a real disciplinary value for the undergraduate student Within this general purpose my aim has been two fold I have sought to show in the first place that psychology is above the laboratory that we employ our instruments of precision not for their own sake but solely because they help us to a refined and more accurate introspection And secondly just as in my Outline of Psychology and Primer of Psychology I gave the results of experimentation a prominent place in the psychological system so here I have treated the selected experiments not as separate exercises but as points of departure for systematic discussion PsycINFO Database Record c 2005 APA all rights reserved **Teachers Manual to Accompany**

Laboratory for General Chemistry Beran, 1986-05 **Instructor's Manual with Test Bank [for] Basic Concepts of Chemistry, Fourth Edition** Alan Sherman, 1988 Instructor's Guide Alan Sherman, Leonard Russikoff, Sharon

Sherman, 1976 **Laboratory Manual for Principles of General Chemistry** Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures *Exploring General Chemistry in the Laboratory* Colleen F. Craig, Kim N.

Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science Pp/Chemistry

BarCharts, Inc., 2008-06-18 Lab Manual for General, Organic, and Biochemistry Denise Guinn, Rebecca

Brewer, 2009-08-21 Teaching all of the necessary concepts within the constraints of a one term chemistry course can be challenging Authors Denise Guinn and Rebecca Brewer have drawn on their 14 years of experience with the one term course to write a textbook that incorporates biochemistry and organic chemistry throughout each chapter emphasizes cases related to allied health and provides students with the practical quantitative skills they will need in their professional lives Essentials of General Organic and Biochemistry captures student interest from day one with a focus on attention getting applications

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Robert J. Artz,1982 **Bulletin of Wake Forest University** Wake Forest College,Wake Forest University,1918

Laboratory Manual for Chemistry Fundamentals Phyllis Buell,James Girard,2002-08 Physical Sciences Parallel Curriculum Units for Science, Grades 6-12 Jann H. Leppien,Jeanne H. Purcell,2011-02-15 Based on the best selling book The Parallel Curriculum this resource deepens teachers understanding of how to use the Parallel Curriculum Model PCM to provide rigorous learning opportunities for students in science grades 6 12 This collection of sample units and lessons within each unit were developed by experienced teachers and demonstrate what high quality curriculum looks like within a PCM framework Ideal for use with high ability students the units revolve around genetics the convergence of science and society the integration of English and Biology and the Periodic Table Lessons include pre and post assessments Resources in Education ,2000-04 Chemistry (Teacher Guide) Dr. Dennis Englin,2018-02-26 This book was created to help teachers as they instruct students through the Master s Class Chemistry course by Master Books The teacher is one who guides students through the subject matter helps each student stay on schedule and be organized and is their source of accountability along the way With that in mind this guide provides additional help through the laboratory exercises as well as lessons quizzes and examinations that are provided along with the answers The lessons in this study emphasize working through procedures and problem solving by learning patterns The vocabulary is kept at the essential level Practice exercises are given with their answers so that the patterns can be used in problem solving These lessons and laboratory exercises are the result of over 30 years of teaching home school high school students and then working with them as they proceed through college Guided labs are provided to enhance instruction of weekly lessons There are many principles and truths given to us in Scripture by the God that created the universe and all of the laws by which it functions It is important to see the hand of God and His principles and wisdom as it plays out in chemistry This course integrates what God has told us in the context of this study Features Each suggested weekly schedule has five easy to manage lessons that combine reading and worksheets Worksheets quizzes and tests are perforated and three hole punched materials are easy to tear out hand out grade and store Adjust the schedule and materials needed to best work within your educational program Space is given for assignments dates There is flexibility in scheduling Adapt the days to your school schedule Workflow Students will read the pages in their book and then complete each section of the teacher guide They should be encouraged to complete as many of the activities and projects as possible as well Tests are given at regular intervals with space to record each grade About the Author DR DENNIS ENGLIN earned his bachelor s from Westmont College his master of science from California State University and his EdD from the University of Southern California He enjoys teaching animal biology vertebrate biology wildlife biology organismic biology

and astronomy at The Master s University His professional memberships include the Creation Research Society the American Fisheries Association Southern California Academy of Sciences Yellowstone Association and Au Sable Institute of Environmental Studies Laboratory Manual of Pharmaceutical Inorganic Chemistry Dr. Prashant D. Aragade ,Dr. Niranjana S. Mahajan,Mr. Nilesh V. Salunkhe,Dr. A. N. Deshpande,2023-10-02 We are pleased to present the Laboratory Manual of Pharmaceutical Inorganic Chemistry This manual is prepared according to the PCI B Pharm course regulations 2014 and is divided into four sections limit tests identification tests purity tests and preparation of inorganic pharmaceuticals The methods of all the experiments are taken from the latest editions of official books such as the Indian European British and US Pharmacopoeia and research papers so that the latest advancements in the methods or apparatus can be incorporated The purpose of pharmaceutical inorganic chemistry practicals is to provide students with hands on experience in understanding and applying the principles of inorganic chemistry to pharmaceutical applications Through these practical sessions students can learn how to prepare analyze and characterize inorganic pharmaceutical compounds which are important in drug development formulations and quality control processes These practicals also help students gain essential laboratory skills such as safely handling chemicals and using various analytical techniques which are crucial for their future careers in the pharmaceutical industry or research This manual is designed for outcome based education and each experiment is arranged in a uniform way with sections for practical significance practical outcomes PrOs mapping with course outcomes theory resources used procedure precautions observations results conclusion references and synopsis questions Each experiment offers an opportunity for students to perform practical work allowing them to gain proficiency in effectively managing equipments handling glasswares chemicals and reagents and writing reports In addition the questions at the end of the experiments help to enhance students knowledge which will be beneficial for them as they pursue higher studies We acknowledge the help and cooperation of various persons in bringing out this manual We are highly indebted to the authors of the books and articles mentioned in the references which were a major source of information for writing this manual We also thank the publishers designers and printers who worked hard to publish this manual in a timely manner We hope that this manual will be helpful to students in understanding concepts principles and procedures We wish you all the best

Environmental Chemistry in the Lab Ruth Ann Murphy,2022-08-31 Environmental Chemistry in the Lab presents a comprehensive approach to modern environmental chemistry laboratory instruction together with a complete experimental experience The laboratory experiments have an introduction for the students to read a pre lab for them to complete before coming to the lab a data sheet to complete during the lab and a post lab which would give them an opportunity to reinforce their understanding of the experiment completed Instructor resources include a list of all equipment and supplies needed for 24 students a lab preparation guide an answer key to all pre lab and post lab questions sample data for remote learners and a suggested rubric for grading the labs Additional features include Tested laboratory exercises with instructor resources for

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Exploring Physical Science in the Laboratory John T. Salinas ,2019-02-01 This full color manual is designed to satisfy the content needs of either a one or two semester introduction to physical science course populated by nonmajors It provides students with the opportunity to explore and make sense of the world around them to develop their skills and knowledge and to learn to think like scientists The material is written in an accessible way providing clearly written procedures a wide variety of exercises from which instructors can choose and real world examples that keep the content engaging Exploring Physical Science in the Laboratory guides students through the mysteries of the observable world and helps them develop a clear understanding of challenging concepts

Teacher's Guide to Using the Next Generation Science Standards With Gifted and Advanced Learners Cheryll M. Adams,Alicia Cotabish,Debbie Dailey,2021-09-23 A Teacher s Guide to Using the Next Generation Science Standards With Gifted and Advanced Learners provides teachers and administrators with practical examples of ways to build comprehensive coherent and rigorous science learning experiences for gifted and advanced students from kindergarten to high school It provides an array of examples across the four domains of science physical sciences Earth and space sciences life sciences and engineering technology and applications of science Each learning experience indicates the performance expectation addressed and includes a sequence of activities implementation examples connections to the CCSS Math and CCSS ELA and formative assessments Chapters on specific instructional and management strategies assessment and professional development suggestions for implementing the standards within the classroom will be helpful for both teachers and administrators

Cambridge IGCSE® Physics Practical Teacher's Guide with CD-ROM Gillian Nightingale,2016-11-24 This edition of our successful series to support the Cambridge IGCSE Physics syllabus 0625 is fully updated for the revised syllabus for first examination from 2016 The Cambridge IGCSE Physics Practical Teacher s Guide complements the Practical Workbook helping teachers to include more practical work in lessons Specific support is provided for each of the carefully designed investigations to save teachers time The Teacher s Guide contains advice about planning investigations guidance about safety considerations differentiated learning suggestions to support students who might be struggling and to stretch the students who are most able as well as answers to all the questions in the Workbook The Teacher s Guide also includes a CD ROM containing model data to be used in instances when an investigation cannot be carried out

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