

Chemistry Spring Final Study Guide

Unit 10: Acids & Bases

Use the following information and complete the structure of Acids & Bases.

Acids	Bases
Unit 10: Acids & Bases Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below. Label the y-axis. Label the x-axis. 1. Indicate the effect of the pressure. 2. Indicate the effect of the volume. 3. Indicate the effect of the temperature. 4. The gas is at 1 atm. 5. The gas is at 2 atm.	Unit 10: Acids & Bases Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below. Label the y-axis. Label the x-axis. 1. Indicate the effect of the pressure. 2. Indicate the effect of the volume. 3. Indicate the effect of the temperature. 4. The gas is at 1 atm. 5. The gas is at 2 atm.

Unit 11: Chemical Equilibrium

Use the following information and complete the structure of Chemical Equilibrium.

Chemical Equilibrium	Chemical Equilibrium
Unit 11: Chemical Equilibrium Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below. Label the y-axis. Label the x-axis. 1. Indicate the effect of the pressure. 2. Indicate the effect of the volume. 3. Indicate the effect of the temperature. 4. The gas is at 1 atm. 5. The gas is at 2 atm.	Unit 11: Chemical Equilibrium Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below. Label the y-axis. Label the x-axis. 1. Indicate the effect of the pressure. 2. Indicate the effect of the volume. 3. Indicate the effect of the temperature. 4. The gas is at 1 atm. 5. The gas is at 2 atm.

Unit 12: Chemical Equilibrium

Use the following information and complete the structure of Chemical Equilibrium.

Chemical Equilibrium	Chemical Equilibrium
Unit 12: Chemical Equilibrium Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below. Label the y-axis. Label the x-axis. 1. Indicate the effect of the pressure. 2. Indicate the effect of the volume. 3. Indicate the effect of the temperature. 4. The gas is at 1 atm. 5. The gas is at 2 atm.	Unit 12: Chemical Equilibrium Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below. Label the y-axis. Label the x-axis. 1. Indicate the effect of the pressure. 2. Indicate the effect of the volume. 3. Indicate the effect of the temperature. 4. The gas is at 1 atm. 5. The gas is at 2 atm.

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Name: _____

Unit 1: Chemical Equilibrium

Draw the best answer from those provided to each question.

Draw the best answer from those provided to each question.

Determine the number of Significant Figures in each of the following numbers.

Mass = 1.23456789 g	Mass = 1.23456789 g
Volume = 1.23456789 L	Volume = 1.23456789 L

Convert the following into Scientific Notation.

Mass = 1.23456789 g	Convert the following into Scientific Notation.
Volume = 1.23456789 L	Mass = 1.23456789 g

Chemical Equilibrium (Round your answers to 3 places after the decimal. Show work to show your work.)

a. Convert the pH value into [H⁺].

b. Convert the [H⁺] value into pH.

Unit 2: Reaction

For the following compounds, calculate the **Atomic Mass** and the **Molecular Mass** of each compound.

H₂O, H₂O₂, H₂SO₄, H₂SO₃, H₂SO₂, H₂SO₁

Unit 3: Reaction

For the following compounds, calculate the **Atomic Mass** and the **Molecular Mass** of each compound.

H₂O, H₂O₂, H₂SO₄, H₂SO₃, H₂SO₂, H₂SO₁

Unit 4: Reaction

For the following compounds, calculate the **Atomic Mass** and the **Molecular Mass** of each compound.

H₂O, H₂O₂, H₂SO₄, H₂SO₃, H₂SO₂, H₂SO₁

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