

2 homework point(s)

Galvanic Cells

Using tabulated standard reduction potentials from your text, calculate the standard cell potential, E°_{cell} (always positive for a galvanic cell), based on the following (unbalanced) reaction:



$$E^\circ_{\text{cell}} = 0.036 \text{ V}$$

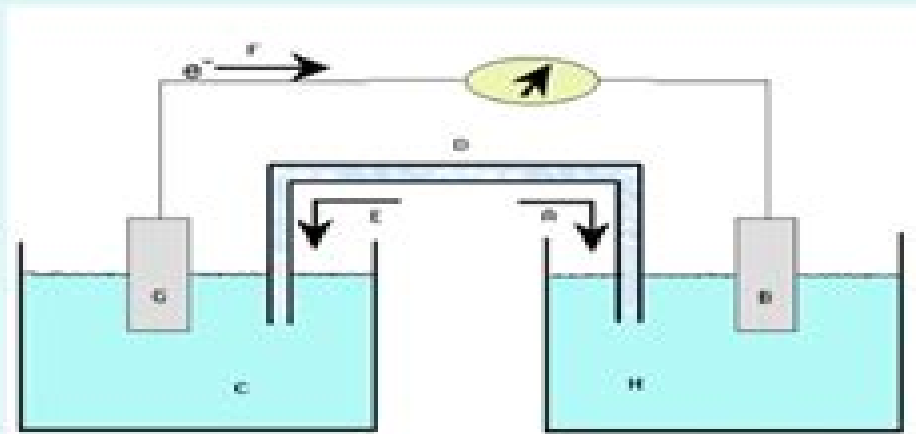
of Table A5.5, p. A24 Zumdahl "Chemical Principles" 7th ed.

1 pt

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A galvanic cell based on the above reaction is constructed according to the generic sketch of a galvanic cell shown below:



Choose the label which best describes the following:

C: Solution containing Cr^{3+} , H^+ and $\text{Cr}_2\text{O}_7^{2-}$

B: $\text{Cl}_2(\text{g})/\text{Pt}$ cathode

The overall, balanced, spontaneous reaction occurring in this (standard) cell is:



False: The salt bridge disrupts the electrical circuit while permitting mixing of the solutions in the anode and cathode compartments.

False: The salt bridge is labeled "D".

True: The solution in tube "D" must contain an electrolyte for the cell to operate.

True: The salt bridge is intended to allow counter ions to flow between the two cell compartments.

False: Tube "D" allows Cr^{3+} ions to migrate readily to the other compartment and to react directly with the chlorine gas.

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Fundamentals of Physics David Halliday, Robert Resnick, Jearl Walker, 2010-03-15 This book arms engineers with the tools to apply key physics concepts in the field A number of the key figures in the new edition are revised to provide a more inviting and informative treatment The figures are broken into component parts with supporting commentary so that they can more readily see the key ideas Material from The Flying Circus is incorporated into the chapter opener puzzlers sample problems examples and end of chapter problems to make the subject more engaging Checkpoints enable them to check their understanding of a question with some reasoning based on the narrative or sample problem they just read Sample Problems also demonstrate how engineers can solve problems with reasoned solutions INCLUDES PARTS 1 4 PART 5 IN

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