

## CH 4. RADICALS

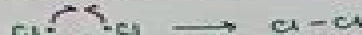
### 1. INITIATION - FORM RADICALS



### 2. PROPAGATION - RADICAL REACTANT → RADICAL PRODUCT



### 3. TERMINATION - 2 RADICALS → NO RADICALS



$$\Delta H_{\text{rxn}} = \sum (\text{bonds broken}) - \sum (\text{bonds formed}) - \text{form bond}$$



$\text{F}_2$  - most reactive  
 $\text{Cl}_2$  - reactive, not selective  
 $\text{Br}_2$  - selective, picky  
 $\text{I}_2$  - not reactive, endo.



RESONANCE + SUBST. ↑ RAD. STABILITY  
 $\text{CH}_3\cdot > \text{R}-\text{C}(\text{H})\cdot > \text{R}-\text{C}(\text{H})_2\cdot > \text{R}-\text{C}(\text{H})_3\cdot$   
 $3^\circ > 2^\circ > 1^\circ > \text{methyl}$

## CH 5. STEREOCHEMISTRY

### CHIRAL - Not symmetrical



### ACHIRAL - SYMMETRY



### MESO - ACHIRAL W/ CHIRAL CENTERS



### CONST. ISOMER - connect. diff.



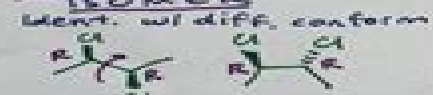
### ENANTIOMER - Invert Chiral centers



### DIASTEROMER - Invert some, not all, chiral centers



### CONFORMATIONAL ISOMER



## CH 6. NUCLEOPHILIC SUBSTITUTION

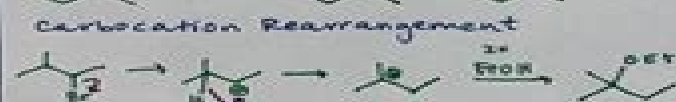


Nuc.  
 Lone pairs  
 (b) charge  
 (b) or  $\delta^+$  charge

LEAVING GROUP:  $\text{I} > \text{Br} > \text{Cl} \gg \text{F or OH}$



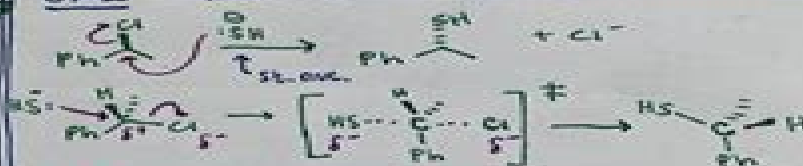
### SN1 → Racemic



H or methyl shift, resonance

\* Weak Nucleophile, Rate =  $k[\text{subst.}]$

### SN2 → Invert chiral center



\* Strong Nucleophile  
 $1^\circ$  or methyl

$$\text{rate} = k[\text{substrate}][\text{nucleophile}]$$

# Guide Quick Notes Reactions Iitjee Organic Chemistry

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