

LIFE

New knowledge about wildlife reveals that
**Zoos Drive
Animals Psycho**

A zoo orangutan
lives and moans



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web solution 1 the two half reactions balanced as if in acidic solution $2\text{CN}^- + \text{Au} \rightarrow \text{Au} + \text{CN}^- + 2\text{e}^-$ $2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2 + 2\text{e}^-$ make electrons equal convert to basic solution $4\text{CN}^- + 2\text{Au} + 2\text{H}_2\text{O} \rightarrow 2\text{Au} + 4\text{CN}^- + 2\text{H}_2\text{O} + \text{O}_2$ multiplied by a factor of 2 2e^- $2\text{H}_2\text{O} + \text{O}_2 \rightarrow 2\text{H}_2\text{O} + 2\text{OH}^-$ 3 the final answer

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is 0 oxidation ni s ni a 2 a q 2 e now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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