

# ACIDS, BASES AND SALTS

<b>Definition of an acid</b> An acid is a compound which when dissolved in water produces hydrogen ions as the only positively charged ions	<b>What causes the acidic properties of acids?</b> The hydrogen ions ( $H^+$ ) cause the acidic properties and these are formed in the presence of water.
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<b>Another term to refer to an acid</b> An acid is called a <u>proton donor</u>	<b>Why an acid is also called a proton donor?</b> It's because an acid provides protons or hydrogen ions ( $H^+$ ) to other substances during the reaction.
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<b>Substance to which an acid provides protons</b> BASE	<b>A base is then called</b> PROTON ACCEPTOR	<b>Why the base is called a proton acceptor?</b> Because the base accepts hydrogen ions from acids	<b>Equation for the reaction between acid and base</b> $H^+_{(aq)} + OH^-_{(aq)} \longrightarrow H_2O_{(l)}$
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Common laboratory acids		These three common laboratory acids are also called	Why these acids are also called mineral acids?
Hydrochloric acid	HCl	MINERAL ACIDS	They are derived from mineral salts i.e. chlorides for HCl, sulphates for $H_2SO_4$ and nitrates for $HNO_3$
Sulphuric acid	$H_2SO_4$		
Nitric acid	$HNO_3$		

Other mineral acids known		Mineral salts from which the acid is derived	Organic acids known	Naturally occurring acids
Sulphurous acid	$H_2SO_3$	Derived from sulphites	Ethanoic acid ( $CH_3COOH$ )	CITRIC ACID from lemons
Carbonic acid	$H_2CO_3$	Derived from carbonates		TARTARIC ACID from grapes
Phosphoric acid	$H_3PO_4$	Derived from phosphates	Methanoic acid ( $HCOOH$ )	ACETIC ACID from vinegar
Nitrous acid	$HNO_2$	Derived from nitrites		LACTIC ACID from sour milk
				Hydrochloric acid from digestive juices

<b>Whenever an acid is dissolved in water, it produces</b> HYDROGEN IONS	<b>Term given to the number of hydrogen ions produced by one molecule of an acid</b> BASICITY OF AN ACID	<b>Definition of basicity of an acid</b> BASICITY of an acid is the <u>number of hydrogen ions</u> produced by <u>one molecule</u> of an acid <u>in aqueous solution</u> .
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# Guide Acids Bases And Salts

**Yi-Tong Ma**



## **Guide Acids Bases And Salts:**

## Reviewing **Guide Acids Bases And Salts**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Guide Acids Bases And Salts**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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### **Table of Contents Guide Acids Bases And Salts**

1. Understanding the eBook Guide Acids Bases And Salts
  - The Rise of Digital Reading Guide Acids Bases And Salts
  - Advantages of eBooks Over Traditional Books
2. Identifying Guide Acids Bases And Salts
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Guide Acids Bases And Salts
  - User-Friendly Interface
4. Exploring eBook Recommendations from Guide Acids Bases And Salts
  - Personalized Recommendations
  - Guide Acids Bases And Salts User Reviews and Ratings

- Guide Acids Bases And Salts and Bestseller Lists
- 5. Accessing Guide Acids Bases And Salts Free and Paid eBooks
  - Guide Acids Bases And Salts Public Domain eBooks
  - Guide Acids Bases And Salts eBook Subscription Services
  - Guide Acids Bases And Salts Budget-Friendly Options
- 6. Navigating Guide Acids Bases And Salts eBook Formats
  - ePub, PDF, MOBI, and More
  - Guide Acids Bases And Salts Compatibility with Devices
  - Guide Acids Bases And Salts Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Guide Acids Bases And Salts
  - Highlighting and Note-Taking Guide Acids Bases And Salts
  - Interactive Elements Guide Acids Bases And Salts
- 8. Staying Engaged with Guide Acids Bases And Salts
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Guide Acids Bases And Salts
- 9. Balancing eBooks and Physical Books Guide Acids Bases And Salts
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Guide Acids Bases And Salts
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Guide Acids Bases And Salts
  - Setting Reading Goals Guide Acids Bases And Salts
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Guide Acids Bases And Salts
  - Fact-Checking eBook Content of Guide Acids Bases And Salts
  - Distinguishing Credible Sources

- 
13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
  14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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