Chapter 9:	Electrons in	Atoms and	the Per	iodic Table

Date:

Section 1: Light and Quantized Energy - Notes

Objectives:

- · Compare the wave and particle natures of light.
- Define a quantum of energy, and explain how it is related to an energy change of matter.
- Contrast continuous electromagnetic spectra and atomic emission spectra.

Unanswered Questions:

- In Rutherford's model, the atom's mass is concentrated in the ______

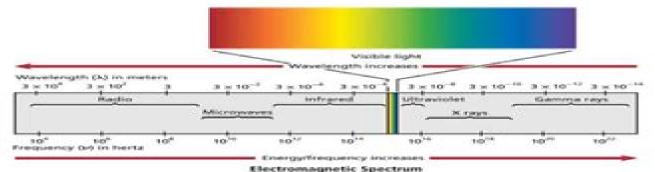
 and _____ move around it.
- This model doesn't explain how the ______ were arranged around the nucleus.
- This model also doesn't explain why ______ charged electrons aren't pulled into the ______ charged nucleus.

The Wave Nature of Light:

_____ when heated in a flame.

- Analysis of this light revealed that an element's ________ in its atoms.

 is related to the arrangement of the _______ in its atoms.
- Visible light is a type of _______.
- The electromagnetic spectrum includes all forms of electromagnetic radiation:



Light And Quantized Energy Study Guide Answer

Margaret G. Kimble

Light And Quantized Energy Study Guide Answer:

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Fundamentals Of Solid-state Electronics: Study Guide Chih Tang Sah,1994-02-08 This companion to Fundamentals of Solid State Electronics provides a helpful summary of the main text for students and lecturers alike The clear typeface large font and point form layout are designed to produce viewgraphs for lectures and to provide ample margins for study notes This Study Guide comes complete with a detailed description of two one semester solid state electronics core courses taught to about 80 100 sophomore junior students each time four years apart It links the contents of the one semester lecture course to the textbook Student Study Guide to accompany Chemistry Martin Silberberg, Libby Weberg, 2005-01-06

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