#### Intermolecular Forces Lab Worksheet



**Directions**: Today you will complete two experiments that compare the properties of water to isopropanol (a type of rubbing alcohol). Before conducting each experiment, you will first read about the experiment and then make a prediction about what you think will happen. Afterward you do the experiment, you will write an explanation of what was happening at the atomic level that let us observe these properties.

Introduction: In many ways, water is a miracle liquid. It is essential for all living things on earth and it is often referred to as a universal solvent because many substances dissolve in it. Water displays unusual properties due to the ways the individual water molecules interact with each other.

 Draw the structural formula (with lone pairs) for the two molecules we will be comparing today, water (H<sub>2</sub>O) and isopropanol (C<sub>2</sub>H<sub>8</sub>O). Draw a dipole arrow to show the direction the electrons are being pulled and label the areas of partial positive and negative charges.

H <sub>2</sub> O	C <sub>1</sub> H <sub>1</sub> O  Note: the bond between C and H is NOT POLAR. Do not draw dipole arrows or partial charges for those bonds.

Which molecule is more polar? \_\_\_\_\_\_\_

#### Experiment #1: Penny

How many drops of water fit on the head of a penny? How many drops of isopropanol fit on the head of a penny?

#### Prediction:

a. I think \_\_\_\_\_ drops of water will fit on the head of the penny because \_\_\_\_\_



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from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement Foundation Course in Chemistry with Case Study Approach for IEE/ NEET/ Olympiad Class 9 - 5th Edition Disha Experts, 2020-07-01 Foundation Chemistry for IIT JEE NEET Olympiad Class 9 is the thoroughly revised and updated 4th edition 2 colour of the comprehensive book for class 9 students who aspire to become Doctors Engineers The book goes for a complete makeover to 2 colour from B W so as to make it more reader friendly The theoretical concepts in the book are accompanied by Illustrations Check Points Do You Know Idea Box and Knowledge Enhancer The book has in total 995 guestions divided into 4 levels of fully solved exercises which are graded as per their level of difficulty Exercise 1 FIB True False Matching Very Short Short and Long Answer Type Questions Exercise 2 Textbook Exemplar and HOTS Questions Exercise 3 4 MCQs 1 Correct MCQs 1 Correct Passage Assertion Reason Multiple Matching and Integer Type Questions The book adheres to the latest syllabus set by the NCERT going beyond by incorporating those topics which will assist the students scale up in the next classes to achieve their academic dreams of Medicine or Engineering These topics are separately highlighted as Connecting Topics and an exercise is developed on the same Study Material Based On NCERT Science Class - IX Dr. Sunita Bhagiya, ,Er. Meera Goyal,2022-02-16 1 Matter In Our Surrounding 2 Is Matter Around us Pure 3 Atoms And Molecules 4 Structure of the atoms 5 The Fundamental Unit of life 6 Tissues 7 Diversity in Living Organisms 8 Motion 9 Force and Laws of Motion 10 Gravitation 11 Work And Energy 12 Sound 13 Why Do we Fall Ill 14 Natural Resources 15 Improvement in Food resources Practical Work Project Work Educational Research and Innovation Fostering Students' Creativity and Critical Thinking What it Means in School Vincent-Lancrin Stéphan, González-Sancho Carlos, Bouckaert Mathias, de Luca Federico, Fernández-Barrerra Meritxell, Jacotin Gwénaël, Urgel Joaquin, Vidal Quentin, 2019-10-24 Creativity and critical thinking are key skills for complex globalised and increasingly digitalised economies and societies While teachers and education policy makers consider creativity and critical thinking as important learning goals it is still unclear to many what it means to develop these skills in a school setting To make it more visible and tangible to practitioners the OECD worked with networks of schools and teachers in 11 countries to develop and trial a set of pedagogical resources that exemplify what it means to teach learn and make progress in creativity and critical thinking in primary and secondary education OAR Quarterly Index of Current Research Results United States. Air Force. Office of Aerospace Research, 1967 **Chemistry** Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues

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