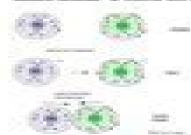
Final Study Guide

Volume Electrons. Electrons on the outcomest shell or orbital, electrons involved to bonding, stome are most stable when they have 8 (this is railed the pate) rate!

Instage: stores with the same atomic number, but different stores messes, because they have different amount of contrars.

Obtaining Numbers—positive or negative number above the symbol that tells us have many electrons an etem.

Some Board a board between a metal and a non-metal where electrons are last and gained



Metal usually loses the electron and the commetal seculty gains in this score that metal to be positively charged and the numberal to be negatively charged. Positives and negatives effects.

Properties :

- Combact electricity when melted or dissolved in water.
- high mediting polisies
- Accordant
- Usually form organiline structures

Considert bounds: a bound between two numericals whose alectrons are shared, both resolution attented to the shared electrons. Electrons can be abased unequally which results in a polar savelant bond. On also form double and tolple bonds.

Propertion:

- Low melting points
- Extend to be suffered
- Usually don't dissolve in water
- Dien't needuct electricity.

The Law of Connervation of state. Mass current be seeded to destroyed. The mass of the products must equal the mass of the resolution.



on you, can use in this picture the come before the reaction is equal to the mass after the reaction.

This rule applies to stome as well.

You must have the same number and type of alone at the businesses of the reaction as you do after the reaction

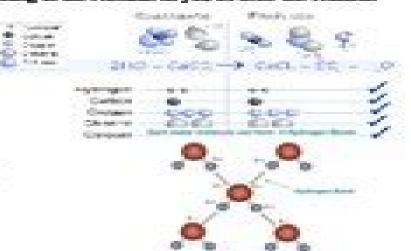
Steps for Reduceting Equations:

- 1. List the status involved in the resultion
- Count how many of each type of etem on each wide.
- Add coefficients to make the number of each tree of atom on each side be equal.
- Recount the number of each type of atom, on each side.
- S. If not equal report steps 3-4.



The second

lie e pedar miderale.



Wilderson (Street)

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