Nume		
TIPIC		

## The Atom and Nuclear Chemistry Review ANSWERS

1.	Neutrons are neutral, but protous arepostive and electrons arenegative					
2.	a) Francism-212 How many neutrons? 125 How many protons? 87					
	b) <sup>64</sup> <sub>17</sub> Co How musty neutrons? _57 How mosty protoss? _27					
	<ul> <li>Give the symbol and hyphenated name for an atom with 20 protons and 22 neutrons</li> </ul>					
	<sup>47</sup> <sub>20</sub> Ca Calcum 42					
3.	What was the first element found to be radioactive?					
4.0	What is happening when an element is radioactive?  The atom is spitting out particles from its mackets.					
5.0	What role do the neutrons play in keeping the nucleus of an atom stable? The stop the protons in the nucleus f from "fighting" (i.e. from repelling each other)					
6.	Atoms with more than88 protons in their madeus are always radioactive.					
700	Define background radiation. The radiation we receive from both natural and human-made sources.					
8	What are some sources of naturally occurring radiation?  Rocks, space, fixed, air					
0.	What are some human-made sources of radiation?  x-rays, TV's, nuclear power plants, nuclear fallout from nuclear weapons testing, nuclear medicine					
10.	What is an alpha particle? A beliann-1 atom					
	Write its symbol. "He					
11.	What is a beta particle? A first moving electron					
	Write its symbol, " pe					
12.	Which is more harmful, an alpha particle or a beta particle? Why? How can each be stopped?					
	An alpha partiale can be stopped by a piece of paper, but to stop a beta particle something thicker like book is moded. Because a beta particle is more penetrating, the beta particle is more dangerous.					
13.	Where do alpha and beta particles come from?the nucleus					
344	When a beta particle forms, a neutron turns into a proton					

## **Ipc Chemistry Review Answers**

**M Woodhall** 

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