

# Iron (Fe) Absorption



**Heme iron**  
**Fe<sup>2+</sup>**



**Non-heme iron**  
**Fe<sup>3+</sup>**

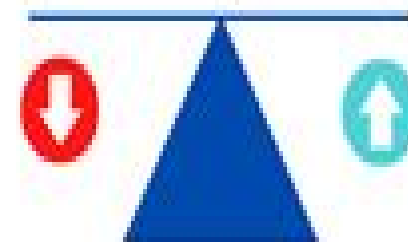
	Heme iron	Non-heme iron
Oxidation state	Fe <sup>2+</sup>	Fe <sup>3+</sup>
Bioavailability	High	Low
Absorption	High	Low

Iron supplement
Bivalent
Ferric fumarate (Fe <sup>2+</sup> )
Ferric gluconate (Fe <sup>2+</sup> )
Ferric sulfate (Fe <sup>2+</sup> )
Ferric glycine sulfate (Fe <sup>2+</sup> )
Trivalent
Iron protein succinylate (Fe <sup>3+</sup> )
Iron polysaccharide complex (Fe <sup>3+</sup> )

Fe<sup>3+</sup> → Fe<sup>2+</sup> → Fe<sup>3+</sup>  
 Die (non- Absorbed Tf-bound  
 heme Fe)

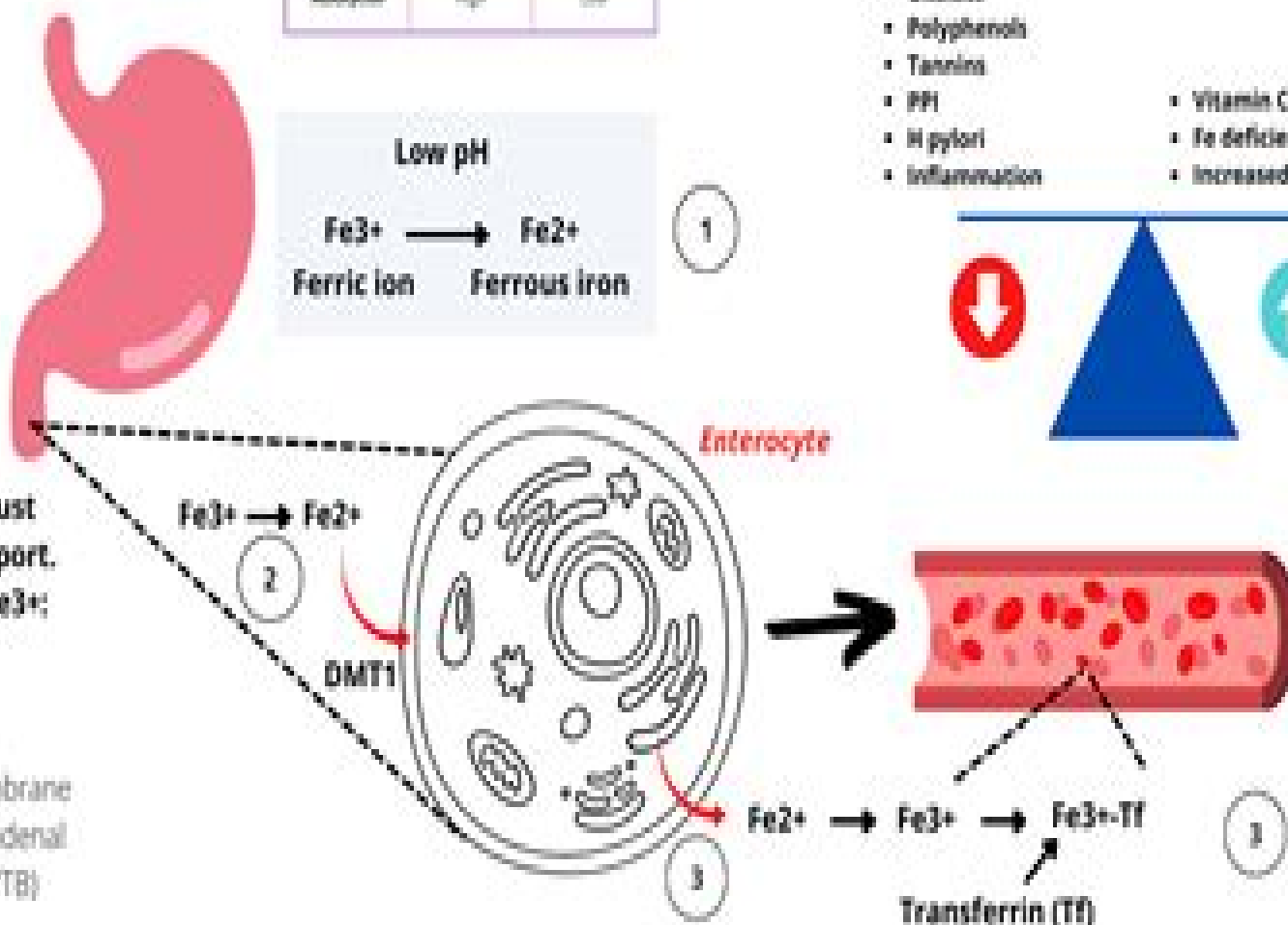
## Nonheme iron absorption

- Phytate
- Oxalate
- Polyphenols
- Tannins
- PH
- H pylori
- Inflammation
- Vitamin C
- Fe deficiency
- Increased erythropoiesis



Ferric iron, therefore, must be reduced before transport.  
 Mechanisms to reduce Fe<sup>3+</sup>:

- 1 Acid pH
- 2 Brush-border membrane ferrireductase, duodenal cytochrome b (DCYTB)



# Iron Absorption Is Impaired By

**IM Harris**



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