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 $3 \text{ } 3 \text{ mg} + 1 \text{ fe}_2\text{o}_3 \rightarrow 2 \text{ fe} + 3 \text{ mgo}$ type of reaction single displacement
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 $\text{naoh} + \text{kno}_3 \rightarrow \text{nano}_3 + \text{koh}$ ch 2 o co 2 h 4 2 2 2o 2 fe 6 nabr 2 febr 6 na 3 caso mg
 $\text{oh} + \text{ca} + \text{oh} + \text{mgso}_4 \rightarrow 2 \text{ } 2 \text{ } 4 \text{ nh}_4\text{oh} + \text{hbr} + \text{h}_2\text{o} + \text{nh}_4\text{br} + \text{pb} + \text{o}_2 + \text{pbo}_2 + \text{na}_2\text{co}_3 + \text{na}_2\text{o} + \text{co}_2 + \text{c} + 4\text{h}_8 + 6\text{o}_4 + \text{co}_2 + 2 \text{ } 4\text{h}_2\text{o} + \text{hcl} + \text{naoh} + \text{h}_2\text{o} + \text{nacl}$

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