



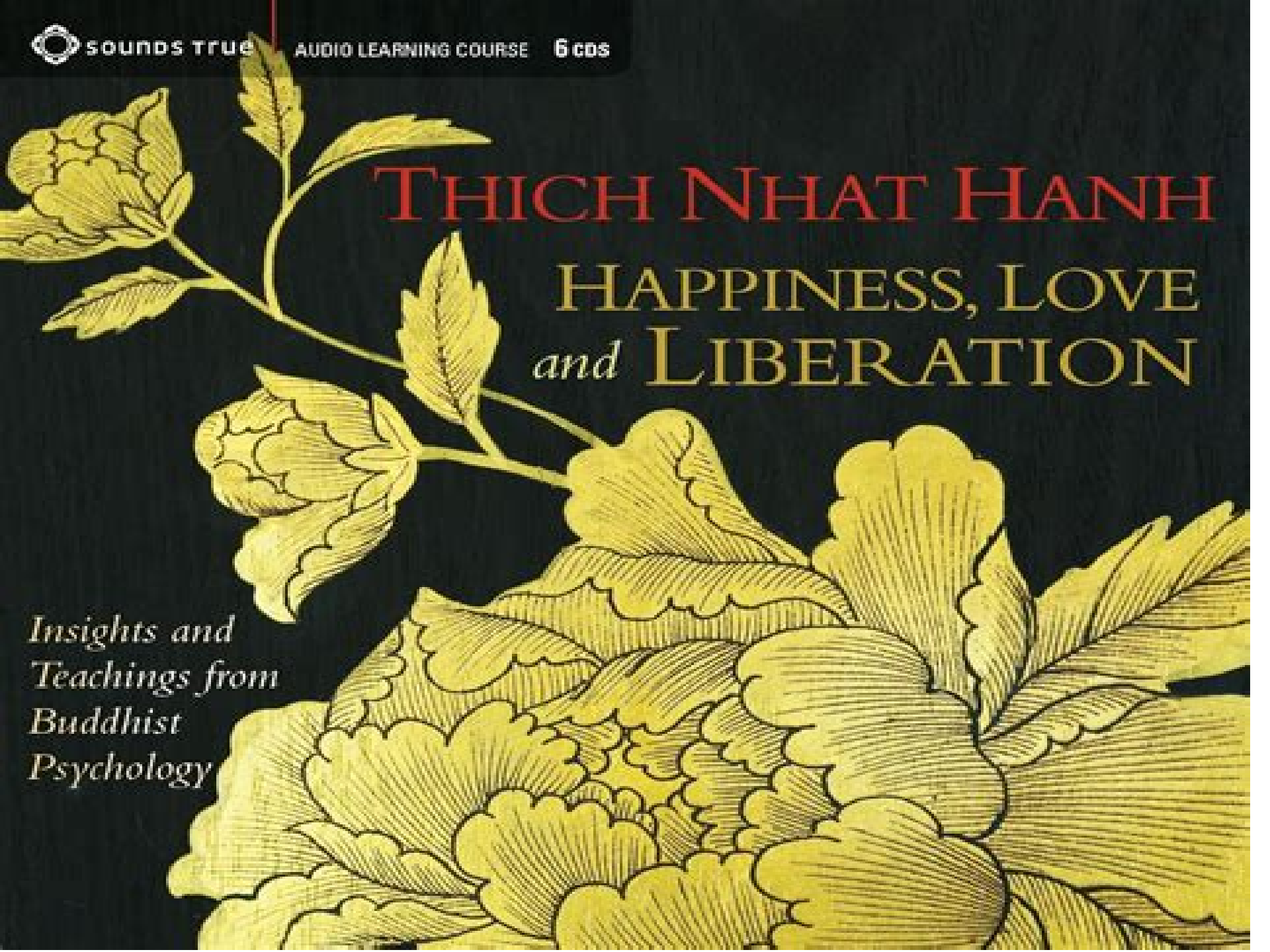
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exothermic endothermic b how much heat will be absorbed or released if 10 g of NO is reacted with 10 g of H_2O

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web answer a 1 explanation the entire reaction was multiplied by 2 therefore ΔH must also be halved 3 answer c explanation

option c is incorrect because 566 kJ/mol is the ΔH per mole of O_2 and not for 2 moles even though options b and d are

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according to the following equation how much heat energy must be expelled by the body to rid the body of one pound of fat

$2\text{C}_3\text{O}_2 + 2\text{SO}_3 \rightarrow 2\text{SO}_2 + 3\text{H}_2\text{O}$ 7914 kJ is this endothermic or exothermic 2

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