

HAZMAT FRA/FRO STUDY GUIDE

The purpose of this study guide is to provide some topic areas that might be useful to focus on in order to be more successful on the AZCFSE written FRA/FRO examination. There is no guarantee that all topics covered on the exam will be listed in this study guide.

1. Types of transportation tanks (Be able to identify and name them. Know modes and responsible parties)
2. Understand DOT placards, labels and markings
3. Understand All DOT hazard classes and divisions
4. Roles of the different agencies: EPA, DOT, Chemtrec, FBI, Postal Service, OSHA
5. When to call Chemtrec and the information they need: what to expect and not to expect
6. Radiation hazards: types, classification, protection, and PPE required
7. PPE levels of protection (protective clothing) & working knowledge of SAR, SCBA, PAR
8. Chemical agents and their classification (DOT, UN, etc)
9. Decontamination (understand the different types)
10. Emergency decontamination procedures
11. Technical decontamination steps
12. Remote shut off's/ remote shut off's on MC carriers (understanding how they work, the different types, and locations)
13. Defensive control techniques (understand and be able to define): absorption, adsorption, damming, diking, dilution, diversion, vapor dispersion, remote valve shutoff
14. Exposure: signs, symptoms, routes, affects, protection, guidelines
15. Review Sheltering-in-place steps
16. Outline three zones
17. Understand the terminology of the Control Zones
18. Command: functions, areas, roles, positions, and safety officer roles
19. Understand the difference between Level I, II, and III Incidents
20. Definitions of permeation, penetration, and degradation
21. Understand how to identify the Table of Initial Isolation and Protective Distances.
22. Emergency Response Guide (ERG): Uses and sections: Understand the operational components of the ERG verses the NIOSH. Understand the isolation zones and color as related to the ERG: Understand the Potential Hazard categories found in the Orange Pages of the ERG
23. Review the steps involved during the protective action phases of an incident
24. Know the definition of a hazardous material as defined by DOT.

Louisiana Hazmat Study Guide

Wolfgang Guggemos

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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web what is a titration a titration is an analytical procedure used to determine the accurate concentration of a sample by reacting it with a standard solution one type of titration uses a neutralization reaction in which an acid and a base react to produce a salt and water

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web donor whereas a base generically represented by b is a proton acceptor when acids and bases combine they swap a proton ha b a hb strong acids e g sulfuric acid have a very pronounced tendency to transfer a proton to a base whereas weak acids

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web understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web jul 19 2023 at the beginning of the titration the solution in the erlenmeyer flask is acidic as the base is added it completely reacts with the acid and the solution in the erlenmeyer flask continues to be acidic but at the equivalence point

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web answer based on the data the strong acid strong base combination of hydrochloric acid and sodium hydroxide had the neutral pH at the equivalence point being 7 this result is expected because a combination of strong acids and strong bases should result in a neutral at the equivalence point

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web may 10 2021 acid base titration lab dataclassroom a titration is a lab technique used to determine the volume of a solution that is needed to react with a given amount of another substance in this activity your goal is to determine the molar concentration and strength of two acid solutions by conducting titrations with a base solution of known

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web spreadsheet computations prelab report this is the type of computations the computer does to determine first and second derivatives refer to these spreadsheet columns for the comments below pH titration graph lab data provide columns a and b

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web 1 approximately 3 g of hydrated oxalic acid $H_2C_2O_4 \cdot 2H_2O$ was weighted to the nearest 0.01 g in a 50 mL beaker 2 about

30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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web solution a titrant volume 0 ml the solution ph is due to the acid ionization of hcl because this is a strong acid the ionization is complete and the hydronium ion molarity is 0 100 m the ph of the solution is then $\text{ph} = -\log 0.100 = 1.000$ b titrant volume 12.50 ml

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web one type of titration uses a neutralization reaction in which an acid and a base react to produce a salt and water in equation 1 the acid is hcl hydrochloric acid and the base is naoh sodium hydroxide when the acid and base react they form nacl sodium chloride which is also known as table salt

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web question lab report acid base titrations purpose of the experiment in this experiment we will titrate an acid with a base

to determine the concentration of an acid with a base to determine the concentration of an acidic sample

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web we can think graphs of absolute value and quadratic functions as transformations of the parent functions x and x^2 importantly we can extend this idea to include transformations of any function whatsoever this fascinating concept allows us to graph many other types of functions like square cube root exponential and logarithmic functions

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web the range of a quadratic function written in standard form $f(x) = a(x - h)^2 + k$ with a positive a value is $f(x) \geq k$ the range of a quadratic function written in standard form with a negative a value is $f(x) \leq k$

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web describing transformations of quadratic functions quadratic function is a function that can be written in the form $f(x) = a(x - h)^2 + k$ where $a \neq 0$ the u shaped graph of a quadratic function is called a parabola in section 1 1 you graphed quadratic functions using tables of values

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web more importantly these equations show the two special forms of these types of functions form equation uses standard form $y = ax^2 + bx + c$ simplified form vertex form $y = a(x - h)^2 + k$ transformations of parent function sketch graphs determine equation from a graph transformation effects when a quadratic function is given in the

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web it can be used as the starting function for transformations on quadratics 1 what is the vertex of the function $f(x) = 2x^2 - 8x + 2$ which statement is true about the function $f(x) = x^2 + 10$

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web when a quadratic function is represented in vertex form following points are to be noted if $h > 0$ graph shifts right by h units if $h < 0$ graph shifts left by h units if $k > 0$ graph shifts upwards by k units if $k < 0$ graph shifts downwards by k units

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