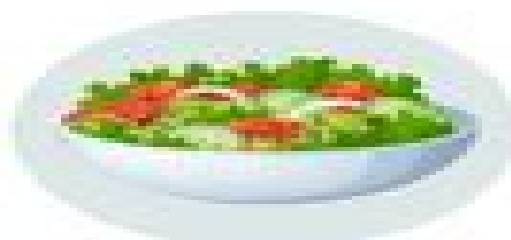


Homogenous vs. Heterogeneous Mixtures Worksheet

What is a Mixture?

A mixture is a combination of two or more substances that are physically combined. This means that the substances are mixed, but they are not chemically bonded and can be separated using physical methods (like filtration, evaporation, etc.).



Reference Guide:

In the table below, we will define both homogeneous and heterogeneous mixtures. We will also give the appearance for each, as well as some examples. Use this reference guide to help remind you of the differences between the two types of mixtures.

Remember: A pure substance consists of only one type of molecule or element and has a fixed composition, while a mixture consists of two or more different substances that are physically combined and can have varying compositions.

Mixture	Definition	Appearance	Examples
Homogeneous	A mixture in which the composition is uniform throughout. Every part of the mixture has the same properties.	Appears the same throughout. You cannot easily see the different components.	<ul style="list-style-type: none"> Saltwater Sugar water Air Vinegar Many alloys (mixtures of metals, like brass)
Heterogeneous	A mixture in which the composition is not uniform throughout. Different parts of the mixture have different properties.	Appears different throughout. You can usually see the different components.	<ul style="list-style-type: none"> Sand and water Oil and water A salad Pasta Soil

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