Name: _

Solubility and Concentration Pretest

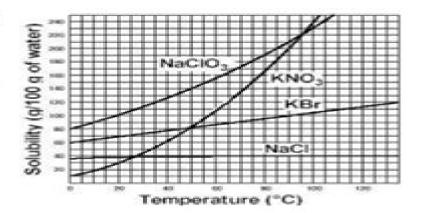
- 1. I mix up a solution of some Kool Aid by using the mix, sugar and water. What is the solute and what is the solvent?
- Define these terms:

Saturated:

Unsaturated:

Supersaturated:

- 3. I add some more sugar to the Kool Aid from #1 and it dissolves. Which of the above three terms best describes the solution? Explain.
- ___ How many grams of KBr are needed to make a saturated solution in 100 g of water at 30 °C?
- In terms of saturation, how would you classify a KBr solution with a concentration of 90 q / 100 g of water at 50 °C?
- 6. _____ If we raised the temperature to 100 °C, what would its saturation be?
- At what temperature is the solubility of KBr the same as that of KNO₁?



Which is more soluble in water, NaClO₃ or KBr? How can you tell?

<u>High School Chemistry Pretest With Answers</u>

Lei Huang

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