

ACID-BASE STUDY GUIDE

- Normal Plasma pH 7.35 to 7.45 (hydrogen (H⁺) concentration)
 - Higher this goes, higher the mortality rate
 - Measure the acidity or alkalinity of blood
 - Homeostatic mechanisms keep pH normal
 - Buffer systems, kidneys & lungs
 - Greater H⁺ concentration = lower pH & more acidic solute
 - Lower H⁺ concentration = higher pH & more alkaline solute
- Major ECF buffer system
 - Bicarbonate-carbonic acid system
 - $\text{CO}_2 + \text{H}_2\text{O} \leftrightarrow \text{H}_2\text{CO}_3 \leftrightarrow \text{H}^+ \text{ and } \text{HCO}_3^-$
 - Kidneys regulate bicarbonate in the ECF
 - Lungs (controlled by medulla) regulate CO₂, thus the carbonic acid in the ECF
 - Lungs are first responders
- Other buffer systems
 - ECF: inorganic phosphates and plasma proteins
 - ICF: proteins, organic, inorganic phosphates
 - Hemoglobin

METABOLIC ACIDOSIS

pH lower than 7.35

- Can occur by a gain of H⁺ or loss of bicarbonate ions (<22 mm Hg)
- Causes
 - Direct loss of bicarbonate (diarrhea, administration of IV nutrition w/o bicarb or bicarb-producing solutes, diuretic use, early renal insufficiency, excess chloride, lower intestinal fistulas, ureterostomies)
 - Most common due to kidney injury
 - Lactic acidosis (higher it is, the worse sepsis is), salicylate poisoning, renal failure, methanol, ethylene, or propylene toxicity, DKA, ketoacidosis due to starvation
 - Systemic issue: diabetes
- Clinical Manifestations
 - Headache, confusion, drowsiness, increased RR/depth, low BP, nausea, vomiting, cold/clammy skin, arrhythmias, shock, peripheral vasodilation, decreased cardiac output
- Medical Management
 - First have to correct underlying issue

Key Acid Base Study Guide

Siddappa N.Byrareddy



Key Acid Base Study Guide:

This book delves into Key Acid Base Study Guide. Key Acid Base Study Guide is a vital topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Key Acid Base Study Guide, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Key Acid Base Study Guide
- Chapter 2: Essential Elements of Key Acid Base Study Guide
- Chapter 3: Key Acid Base Study Guide in Everyday Life
- Chapter 4: Key Acid Base Study Guide in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Key Acid Base Study Guide. This chapter will explore what Key Acid Base Study Guide is, why Key Acid Base Study Guide is vital, and how to effectively learn about Key Acid Base Study Guide.
3. In chapter 2, the author will delve into the foundational concepts of Key Acid Base Study Guide. The second chapter will elucidate the essential principles that need to be understood to grasp Key Acid Base Study Guide in its entirety.
4. In chapter 3, the author will examine the practical applications of Key Acid Base Study Guide in daily life. This chapter will showcase real-world examples of how Key Acid Base Study Guide can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Key Acid Base Study Guide in specific contexts. The fourth chapter will explore how Key Acid Base Study Guide is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Key Acid Base Study Guide. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Key Acid Base Study Guide.

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