Experiment #14:

ISOPENTYL ACETATE

Data Table for Exp #14

Name	Mass (g)
Mass of empty SmL conical vial	25.021 g
Mass of 5mL conical vial + Isopentyl Alcohol	25.834 g
Mass of Isopentyl Alcohol	0.813 g
Mass of Isopentyl Acetate (Theoretical)	1.20 g
Mass of 5mL conical vial + Isopentyl Acetate (collected)	25.943 g
Mass of Isopentyl Acetate (Experiment)	0.922 g
% Yield of Isopentyl Acetate	76.83%
Boiling Point of Isopentyl Acetate	140°C

Calculation:

Theoretical Yield of Isoperate Acetate

9.22*10'3 mol of Isoperityl Alcohol * \frac{1 mol of Isoperityl A cetate}{1 mol Isoperityl Alcohol} * \frac{130.19 g Isoperityl A cetate}{1 mol of Isoperityl A cetate} = 1.20g Isoperityl Acetate (Theoretical)

96 Tield of Isopentyl Acetate =
$$\frac{\text{experiment result}}{\text{theoretical result}} = \frac{0.922 q}{1.20 g} = 76.83\%$$

Isopentyl Acetate Lab Report

A Gutmann

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