

Isotopes Practice Set

- What does atomic number tell you about an element?
- What does the number next to isotopes signify?
- How can you tell isotopes apart?

For each of the following isotopes, write the number of protons, neutrons, and electrons.

	Chlorine-35	Chlorine-37
# of protons	17	17
# of neutrons	18	20
# of electrons	17	17

	Nitrogen-14	Nitrogen-15
# of protons	7	7
# of neutrons	7	8
# of electrons	7	7

	Carbon-12	Carbon-13
# of protons	6	6
# of neutrons	6	7
# of electrons	6	6

	Oxygen-16	Oxygen-18
# of protons	8	8
# of electrons	8	8
# of neutrons	8	10

Fill in the isotopes names including atomic numbers, including atomic numbers from the chart. Use your periodic table and the information provided.

	Isotope: $^{35}_{17}\text{Cl}$	Isotope: $^{37}_{17}\text{Cl}$
# of protons	17	17
# of neutrons	18	20
# of electrons	17	17

	Isotope: $^{14}_{7}\text{N}$	Isotope: $^{15}_{7}\text{N}$
# of protons	7	7
# of neutrons	7	8
# of electrons	7	7

	Isotope: $^{12}_{6}\text{C}$	Isotope: $^{13}_{6}\text{C}$
# of protons	6	6
# of neutrons	6	7
# of electrons	6	6

	Isotope: $^{16}_{8}\text{O}$	Isotope: $^{18}_{8}\text{O}$
# of protons	8	8
# of neutrons	8	10
# of electrons	8	8

Isotopes Practice Set Answer Key

J Ma



Isotopes Practice Set Answer Key:

Isotopes Practice Set Answer Key Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Isotopes Practice Set Answer Key**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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