

# Chemistry Study Guide

### Periodic Table of the Elements

<div><div>H</div><div>1</div><div>1.008</div></div>																	<div><div>He</div><div>2</div><div>4.003</div></div>						
<div><div>Li</div><div>3</div><div>6.941</div></div>	<div><div>Be</div><div>4</div><div>9.012</div></div>																	<div><div>B</div><div>5</div><div>10.811</div></div>	<div><div>C</div><div>6</div><div>12.011</div></div>	<div><div>N</div><div>7</div><div>14.007</div></div>	<div><div>O</div><div>8</div><div>15.999</div></div>	<div><div>F</div><div>9</div><div>18.998</div></div>	<div><div>Ne</div><div>10</div><div>20.180</div></div>
<div><div>Na</div><div>11</div><div>22.990</div></div>	<div><div>Mg</div><div>12</div><div>24.305</div></div>																	<div><div>Al</div><div>13</div><div>26.982</div></div>	<div><div>Si</div><div>14</div><div>28.086</div></div>	<div><div>P</div><div>15</div><div>30.974</div></div>	<div><div>S</div><div>16</div><div>32.06</div></div>	<div><div>Cl</div><div>17</div><div>35.45</div></div>	<div><div>Ar</div><div>18</div><div>39.948</div></div>
<div><div>K</div><div>19</div><div>39.098</div></div>	<div><div>Ca</div><div>20</div><div>40.078</div></div>	<div><div>Sc</div><div>21</div><div>44.956</div></div>	<div><div>Ti</div><div>22</div><div>47.88</div></div>	<div><div>V</div><div>23</div><div>50.942</div></div>	<div><div>Cr</div><div>24</div><div>51.996</div></div>	<div><div>Mn</div><div>25</div><div>54.938</div></div>	<div><div>Fe</div><div>26</div><div>55.845</div></div>	<div><div>Co</div><div>27</div><div>58.933</div></div>	<div><div>Ni</div><div>28</div><div>58.69</div></div>	<div><div>Cu</div><div>29</div><div>63.546</div></div>	<div><div>Zn</div><div>30</div><div>65.38</div></div>	<div><div>Ga</div><div>31</div><div>69.723</div></div>	<div><div>Ge</div><div>32</div><div>72.63</div></div>	<div><div>As</div><div>33</div><div>74.922</div></div>	<div><div>Se</div><div>34</div><div>78.96</div></div>	<div><div>Br</div><div>35</div><div>79.904</div></div>	<div><div>Kr</div><div>36</div><div>83.80</div></div>						
<div><div>Rb</div><div>37</div><div>85.468</div></div>	<div><div>Sr</div><div>38</div><div>87.62</div></div>	<div><div>Y</div><div>39</div><div>88.906</div></div>	<div><div>Zr</div><div>40</div><div>91.224</div></div>	<div><div>Nb</div><div>41</div><div>92.906</div></div>	<div><div>Mo</div><div>42</div><div>95.94</div></div>	<div><div>Tc</div><div>43</div><div>98.906</div></div>	<div><div>Ru</div><div>44</div><div>101.07</div></div>	<div><div>Rh</div><div>45</div><div>102.91</div></div>	<div><div>Pd</div><div>46</div><div>106.42</div></div>	<div><div>Ag</div><div>47</div><div>107.87</div></div>	<div><div>Cd</div><div>48</div><div>112.41</div></div>	<div><div>In</div><div>49</div><div>114.82</div></div>	<div><div>Sn</div><div>50</div><div>118.71</div></div>	<div><div>Sb</div><div>51</div><div>121.76</div></div>	<div><div>Te</div><div>52</div><div>127.6</div></div>	<div><div>I</div><div>53</div><div>126.91</div></div>	<div><div>Xe</div><div>54</div><div>131.29</div></div>						
<div><div>Cs</div><div>55</div><div>132.91</div></div>	<div><div>Ba</div><div>56</div><div>137.33</div></div>																	<div><div>Tl</div><div>81</div><div>204.38</div></div>	<div><div>Pb</div><div>82</div><div>207.2</div></div>	<div><div>Bi</div><div>83</div><div>208.98</div></div>	<div><div>Po</div><div>84</div><div>209</div></div>	<div><div>At</div><div>85</div><div>210</div></div>	<div><div>Rn</div><div>86</div><div>222</div></div>
<div><div>Fr</div><div>87</div><div>223</div></div>	<div><div>Ra</div><div>88</div><div>226</div></div>																	<div><div>Uut</div><div>113</div><div>288</div></div>	<div><div>Fl</div><div>114</div><div>289</div></div>	<div><div>Uup</div><div>115</div><div>288</div></div>	<div><div>Lv</div><div>116</div><div>293</div></div>	<div><div>Uus</div><div>117</div><div>294</div></div>	<div><div>Uuo</div><div>118</div><div>294</div></div>
		<div><div>Lanthanide series</div><div>LaCePrNdPmSmEuGdTbDyHoErTmYbLu</div></div>																					
		<div><div>Actinide series</div><div>AcThPaUaNpPuAmCmBkCfEsFmMdNoLr</div></div>																					

## Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles ✎

## Matter:

Anything that has **mass** and **occupies space** ✎

Examples: air, gases, liquids, solids!

## Atoms:

Atoms are the **smallest, indivisible particle** ✎

- **Outer part** → *shell/orbit/energy level* → carries the electrons ( $e^-$ )
- **Inner part** → *nucleus* → houses the protons ( $p^+$ ) and neutrons ( $n$ )
- Outer & Inner part → together these are the **subatomic particles**

## Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	$p^+$	1	+1	Nucleus
Neutron	$n$	1	0	Nucleus
Electron	$e^-$	$\sim 0$	-1	Orbit/Shell

# Keystone Chemistry Study Guide Answers

**M Planty**



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## **Table of Contents Keystone Chemistry Study Guide Answers**

1. Understanding the eBook Keystone Chemistry Study Guide Answers
  - The Rise of Digital Reading Keystone Chemistry Study Guide Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Keystone Chemistry Study Guide Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Keystone Chemistry Study Guide Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Keystone Chemistry Study Guide Answers
  - Personalized Recommendations
  - Keystone Chemistry Study Guide Answers User Reviews and Ratings
  - Keystone Chemistry Study Guide Answers and Bestseller Lists
5. Accessing Keystone Chemistry Study Guide Answers Free and Paid eBooks
  - Keystone Chemistry Study Guide Answers Public Domain eBooks
  - Keystone Chemistry Study Guide Answers eBook Subscription Services
  - Keystone Chemistry Study Guide Answers Budget-Friendly Options

6. Navigating Keystone Chemistry Study Guide Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Keystone Chemistry Study Guide Answers Compatibility with Devices
  - Keystone Chemistry Study Guide Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Keystone Chemistry Study Guide Answers
  - Highlighting and Note-Taking Keystone Chemistry Study Guide Answers
  - Interactive Elements Keystone Chemistry Study Guide Answers
8. Staying Engaged with Keystone Chemistry Study Guide Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Keystone Chemistry Study Guide Answers
9. Balancing eBooks and Physical Books Keystone Chemistry Study Guide Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Keystone Chemistry Study Guide Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Keystone Chemistry Study Guide Answers
  - Setting Reading Goals Keystone Chemistry Study Guide Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Keystone Chemistry Study Guide Answers
  - Fact-Checking eBook Content of Keystone Chemistry Study Guide Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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employee excels at and where they need to improve i separate my assessments into two sections which contributions represent the employee s best work constructive feedback on areas of growth

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