

## CH 4. RADICALS

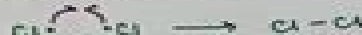
### 1. INITIATION - FORM RADICALS



### 2. PROPAGATION - RADICAL REACTANT → RADICAL PRODUCT



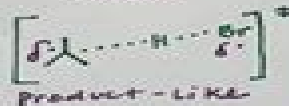
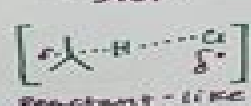
### 3. TERMINATION - 2 RADICALS → NO RADICALS



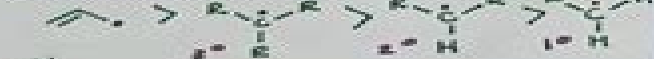
$$\Delta H_{\text{rxn}} = \sum (\text{bonds broken}) - \sum (\text{bonds formed}) - \text{form bond}$$



$\text{F}_2$  - most reactive  
 $\text{Cl}_2$  - reactive, not selective  
 $\text{Br}_2$  - selective, picky  
 $\text{I}_2$  - not reactive, endo.



RESONANCE + SUBST. ↑ RAD. STABILITY

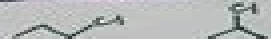


## CH 5. STEREOCHEMISTRY

### CHIRAL - Not symmetrical



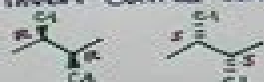
### CONST. ISOMER - connect. diff.



### ACHIRAL - SYMMETRY



### ENANTIOMER - Invert chiral centers



### DIASTEROMER - Invert some, not all, chiral centers



### MESO - ACHIRAL W/ CHIRAL CENTERS



### CONFORMATIONAL ISOMER

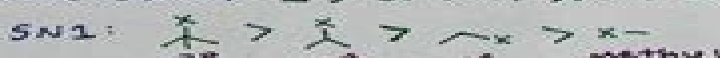


## CH 6. NUCLEOPHILIC SUBSTITUTION

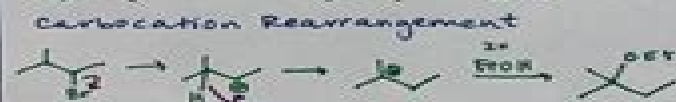


**Nuc.**  
 Lone pairs  
 (b) charge  
 (b) or  $\delta^-$  charge

LEAVING GROUP:  $\text{I} > \text{Br} > \text{Cl} \gg \text{F}$  or  $\text{OH}$

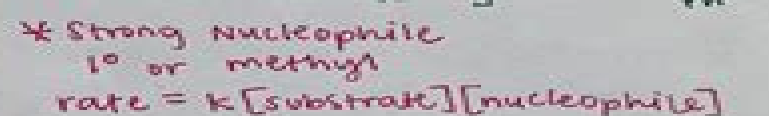
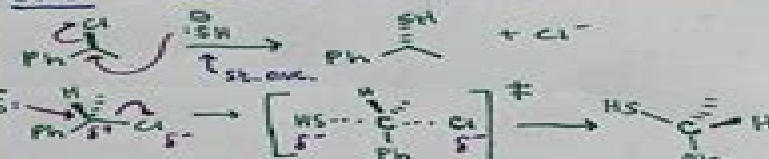


### SN1 → Racemic



\* Weak Nucleophile, Rate =  $k[\text{subst.}]$

### SN2 → Invert chiral center



\* Strong Nucleophile  
 $1^\circ$  or methyl  
 rate =  $k[\text{substrate}][\text{nucleophile}]$

# Guide Quick Notes Reactions Iitjee Organic Chemistry

**Jessica J Manson**



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