

INSTRUCTOR'S SOLUTIONS MANUAL

MATHEMATICAL PROOFS A TRANSITION TO ADVANCED MATHEMATICS FOURTH EDITION

Gary Chartrand

Western Michigan University

Albert D. Polimeni

State University of New York at Fredonia

Ping Zhang

Western Michigan University

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$\text{Mg}^{2+} + 2\text{OH}^{-} + 2\text{H}^{+} + 2\text{NO}_3^{-} + 2\text{H}_2\text{O} + \text{Mg}^{2+} + 2\text{NO}_3^{-}$ net ionic equation
balanced total equation $\text{H}_2\text{SO}_4(\text{aq}) + 2\text{LiOH}(\text{aq}) \rightarrow \text{H}_2\text{O}$