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CONNECTING MIND, RESEARCH,
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web you may have to consult the solubility rules write the net ionic equation for the reaction of $\text{BaCl}_2(\text{aq})$ and $\text{Na}_2\text{SO}_4(\text{aq})$ you may have to consult the solubility rules write

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web choose the correct net ionic equation representing the reaction that occurs when solutions of potassium carbonate and copper i chloride are mixed a $\text{K}_2\text{CO}_3 \text{ aq}$ $2\text{CuCl} \text{ aq}$

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web the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in

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web the net ionic equation for equation 1 is given as equation 3 $\text{s}_2 \text{ aq}$ $\text{Fe}_2 \text{ aq}$ gt fes s eq 3 8 note that the phase indicator aq is optional on ionic species but is

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web balanced chemical equation $2 \text{NaOH (aq)} + \text{MgCl}_2 \text{ (aq)} \rightarrow 2 \text{NaCl (aq)} + \text{Mg(OH)}_2 \text{ (s)}$ full ionic equation $2 \text{Na}^+ \text{ (aq)} + 2 \text{OH}^- \text{ (aq)} + \text{Mg}^{2+} \text{ (aq)} + 2 \text{Cl}^- \text{ (aq)} \rightarrow 2 \text{Na}^+ \text{ (aq)} + 2 \text{Cl}^- \text{ (aq)} + \text{Mg(OH)}_2 \text{ (s)}$