Name: \_\_\_ Date: Mods:

## Honors Chemistry Midterm: Review Packet

#### PART 1

Write in the number of significant figures for each of the following measurements:

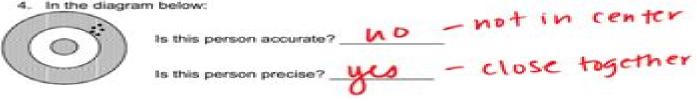
3 0.00290 g 4 50.00 m 2 0.070 kg /

- Using the following masses: 0.0562 kg, 124.213 g and 1635 mg.
  The answer should be reported as: \_\_\_\_\_\_g The answer should be reported as: \_\_\_\_\_\_g 1.635g
- 3. a) A 5.75 mL sample of mercury has a measured mass of 77.05 g. The density is 13.49/mL

b) Mercury's accepted density is 13.53 g/mL. The % error in your measurement is: . 961 70

acture ×100 113.639/ms -13.49/ms ×100

In the diagram below:



- 5. Identify the following as a chemical change (CC), physical change (PC), chemical property (CP), or physical property (PP):
  - a) burning wood CC

b) mass of wood PP

c) flammability of wood CP d) chopping up wood PC

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**University of California (System)** 

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