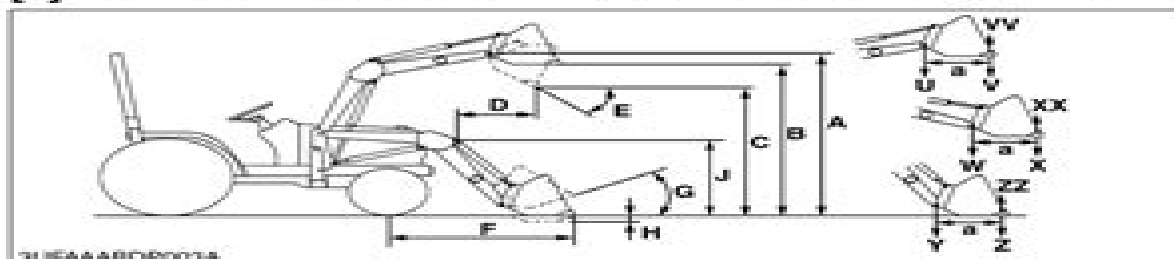


[3] DIMENSIONAL AND OPERATIONAL SPECIFICATIONS



■ Dimensional Specifications

Loader Model		LA1403	
Tractor Model		M96S, M108S	
Boom Cylinder Fulcrum Point		HEIGHT	POWER
(A)	Maximum lift height to pivot pin	3700 mm (145.7 in.)	3370 mm (132.7 in.)
(B)	Maximum lift height under level bucket	3470 mm (136.6 in.)	3160 mm (124.4 in.)
(C)	Clearance with bucket dumped	2660 mm (104.8 in.)	2490 mm (98.0 in.)
(D)	Reach at maximum lift height (Dumping Reach)	590 mm (23.2 in.)	975 mm (38.4 in.)
(E)	Maximum dump angle	0.91 rad (52 °)	1.1 rad (63 °)
(F)	Reach with bucket on ground	2230 mm (87.8 in.)	
(G)	Bucket roll-back angle	0.70 rad (40 °)	
(H)	Digging depth	60 mm (2.4 in.)	65 mm (2.6 in.)
(J)	Overall height in carry position	1700 mm (66.9 in.)	
(a)	Length	600 mm (31.5 in.)	

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■ Operational Specifications

Loader Model		LA1403	
Tractor Model		M96S, M108S	
Boom Cylinder Fulcrum Point		HEIGHT	POWER
Lift capacity (Bucket bottom mid point)		1350 kg (2976 lbs)	1400 kg (3086 lbs)
(U)	Lift capacity (Bucket pivot pin, max. height)	1885 kg (4176 lbs)	1950 kg (4299 lbs)
(V)	Lift capacity (800 mm forward, max. height)	1210 kg (2668 lbs)	1350 kg (2976 lbs)
(W)	Lift capacity (Bucket pivot pin, 1500 mm height)	2125 kg (4685 lbs)	2380 kg (5247 lbs)
(X)	Lift capacity (800 mm forward, 1500 mm height)	1575 kg (3472 lbs)	1640 kg (3615 lbs)
(Y)	Breakout force (Bucket pivot pin)	24860 N (5589 lbf)	26250 N (5876 lbf)
(Z)	Breakout force (800 mm forward)	17190 N (3864 lbf)	22200 N (4991 lbf)
(VV)	Bucket roll-back force at maximum height	13720 N (3084 lbf)	16080 N (3615 lbf)
(XX)	Bucket roll-back force at 1.5 m (59 in.)	25860 N (5814 lbf)	
(ZZ)	Bucket roll-back force at ground level	25810 N (5802 lbf)	
Raising time		4.7 sec.	
Lowering time		3.7 sec.	
Bucket dumping time		2.8 sec.	
Bucket roll-back time		2.9 sec.	

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Kubota La1403 Front Loader Workshop Service Repair Manual

Raffaela Di Napoli



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web k sp of calcium hydroxide 1 author j m mccormick last update october 13 2013 introduction an equilibrium constant K is related to ΔG for a process through eqn 1 and as long as ΔH and ΔS are independent of temperature eqn 1 can be combined with eqn 2 to determine ΔH and ΔS from the temperature dependence of K

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$ $5 \times 10^{-2} \text{ M}$ calcium ions $5 \times 10^{-2} \text{ M}$ hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

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web the chemical equation used was $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$ the solubility product was determined to be ksp Ca(OH)_2 from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

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web jul 12 2023 because the concentration of a pure solid such as Ca_3PO_4 is a constant it does not appear explicitly in

the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore
$$\text{Ca}_3(\text{PO}_4)_2 \rightleftharpoons 3\text{Ca}^{2+} + 2\text{PO}_4^{3-}$$
$$K_{\text{sp}} = [\text{Ca}^{2+}]^3 [\text{PO}_4^{3-}]^2$$

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web therefore it was determined through error calculations and ease of use that the most reliable valid method to use to get the closest ksp value to the theoretical ksp for $\text{Ca}(\text{OH})_2$ was the determination through ph method 2 introduction calcium hydroxide is a strong base soluble in water

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web we will filter the solid calcium hydroxide from the solution and then titrate the dissolved calcium hydroxide in solution with a known concentration of hydrochloric acid using titrimetric calculations we will determine the Ca^{2+} and OH^- concentrations in solution

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web calculate the ksp for calcium hydroxide Ca(OH)_2 $K_{sp} = 5 \times 10^{-4}$ find the accepted value of the ksp for calcium hydroxide and compare it with your value discuss the discrepancy and suggest possible sources of experimental error accepted value for ksp for sodium hydroxide is 5×10^{-6}

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