

Name _____ Date _____ Class _____

CHAPTER 4 STUDY GUIDE FOR CONTENT MASTERY**The Structure of the Atom****Section 4.1 Early Theories of Matter***In your textbook, read about the philosophers, John Dalton, and defining the atom.**For each statement below, write true or false.*

- false** _____ 1. Ancient philosophers regularly performed controlled experiments.
- true** _____ 2. Philosophers formulated explanations about the nature of matter based on their own experiences.
- true** _____ 3. Both Democritus and Dalton suggested that matter is made up of atoms.
- true** _____ 4. Dalton's atomic theory stated that atoms separate, combine, or rearrange in chemical reactions.
- false** _____ 5. Dalton's atomic theory stated that matter is mostly empty space.
- false** _____ 6. Dalton was correct in thinking that atoms could not be divided into smaller particles.
- true** _____ 7. Dalton's atomic theory stated that atoms of different elements combine in simple whole-number ratios to form compounds.
- true** _____ 8. Dalton thought that all atoms of a specific element have the same mass.
- false** _____ 9. Democritus proposed that atoms are held together by chemical bonds, but no one believed him.
- true** _____ 10. Dalton's atomic theory was based on careful measurements and extensive research.
- false** _____ 11. There are no instruments powerful enough to measure atoms so that they can be seen.
- true** _____ 12. The smallest particle of an element that retains the properties of that element is called an atom.

Name _____ Date _____ Class _____

CHAPTER 4 STUDY GUIDE FOR CONTENT MASTERY**Section 4.2 Subatomic Particles and the Nuclear Atom***In your textbook, read about discovering the electron and the nuclear atom.**For each item in Column A, write the letter of the matching item in Column B.*

Column A	Column B
5. _____ 1. Proposed the nuclear atomic model	a. Thomson
6. _____ 2. Determined the mass-to-charge ratio of an electron	b. Millikan
7. _____ 3. Calculated the mass of an electron	c. Rutherford

Draw and label a diagram of each atomic model.

4. **plum pudding model**

Drawing should look like a ball of chocolate chip cookie dough. The chocolate chips should be labeled with negative charge or as electrons. The dough should be labeled as evenly distributed positive charges.

5. **no-lose atomic model**

Drawing should look like a peach with a pit. The pit should be labeled nucleus and should include labeled protons and neutrons. The outer circle of the peach should be labeled electrons.

*In your textbook, read about the discovery of protons and neutrons.**Complete the following table of protons, electrons, and neutrons characteristics.*

Particle	Symbol	Location	Relative Charge	Relative Mass
1. Proton	p^+	In the nucleus	$1+$	1
2. Neutron	n^0	In the nucleus	0	1
3. Electron	e^-	In the space surrounding the nucleus	$1-$	$1/1840$

It Essentials Chapter 4 Study Guide Answers

**William Panek,Tylor Wentworth,James
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CCNP: Building Scalable Cisco Internetworks Study Guide Carl Timm,Wade Edwards,2006-02-20 Here s the book you need to prepare for Cisco s Building Scalable Cisco Internetworks BSCI exam 642 801 This Study Guide provides In depth coverage of key exam topics Practical information on designing and implementing scalable Cisco internetworks Hundreds of challenging review questions Leading edge exam preparation software including a test engine and electronic flashcards Authoritative coverage of all exam objectives including Using classful classless distance vector and link state routing protocols Using VLSM to extend IP addresses Configuring EIGRP OSPF BGP and IS IS environments Configuring and verifying router redistribution in a network Configuring policy based routing using route maps Utilizing the three layer hierarchical design model Identifying IP addressing schemes including features of IPv6 Verifying OSPF operation in a single and multiple areas Ensuring proper operation of Integrated IS IS on Cisco routers Interpreting the output of various show and debug commands Note CD ROM DVD and other supplementary materials are not included as part of eBook file

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