

Gas Stoichiometry Exam
2016-2017

Name KLAMY KEY
Period _____ Date _____

Multiple Choice

Refer to the set of graphs below when answering the next two questions:

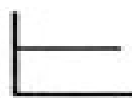
Graph #1



Graph #2



Graph #3



Graph #4



- B 1. Which of the graphs above is the best qualitative graph of pressure versus volume of nitrogen at constant temperature and number of moles?

A) Graph #1
B) Graph #2
C) Graph #3
D) Graph #4

→ INVERSE

- A 2. Which of the graphs above is the best qualitative graph of Kelvin temperature versus pressure of an ideal gas at constant volume and number of moles?

A) Graph #1
B) Graph #2
C) Graph #3
D) Graph #4

→ DIRECT

- D 3. Use the kinetic molecular theory of gases to predict what would happen to a closed sample of a gas whose temperature increased while its volume decreased.

→ MOVE FASTER, MORE COLLISIONS
A) The average kinetic energy of the molecules of the gas would decrease.
B) Its pressure would decrease.
C) The number of moles of gas would decrease.
D) Its pressure would increase.

- B 4. A hydrocarbon gas with the empirical formula CH_2 has a density of 1.3 g/L at 0°C and 1.00 atm. A possible formula for the hydrocarbon is:

A) CH_2
B) C_2H_4
C) C_3H_6
D) C_4H_8
E) C_5H_{10}

~~1.3 g/L~~

$$1.3 \text{ g/L} \times 22.4 \text{ L/mol} \approx 30 \text{ g/mol}$$

Holt Modern Chemistry Gases Test A Answers

William L. Ramsey



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The United States Catalog ,1904 Books in Print Supplement ,2002 *Essentials of Modern Chemistry* Charles Elwood Dull,1918 Holt Chemistry Ralph Thomas Myers,2004 *The American Bookseller* ,1904 **Modern Chemistry** Nicholas D. Tzimopoulos,1993 The Publishers Weekly ,1908 **Lab Experiments Modern Chemistry** Harold Clark Metcalfe,1982 Holt General Science William L. Ramsey,1988 **Modern Chemistry** Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2001 **Adult Catalog: Subjects** Los Angeles County Public Library,1970 *The Cumulative Book Index* ,1998 A world list of books in the English language **El-Hi Textbooks & Serials in Print, 2005** ,2005

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