

Step 2: Calculate the ages of the metamorphic and igneous layers based on the following information.

- A mineral sample from rock unit B has 150,000 atoms of uranium-235 and 50,000 atoms of lead-207.
- A mineral sample from rock unit C has 180,000 atoms of uranium-235 and 20,000 atoms of lead-207.
- A mineral sample from rock unit G has 100,000 atoms of potassium-40 and 100,000 atoms of argon-40.

Use **Figure 9.8**, if necessary, to help you determine calculate how many half-lives have passed since rock units B, C, and G were each formed, based on the remaining amount of a parent isotope compared to the original amount of parent isotope.

Recall that the original amount of parent isotope can be calculated by adding together the amount of parent isotope plus the amount of daughter isotope, making the assumption that neither has escaped the sample. Also note that if less time has passed than a full half-life, you must calculate what portion of a half-life has passed to make the calculations.

Once you know how many half-lives have past since the sample crystallized, use the half-life table in **Figure 9.9** to calculate an absolute date in mya (million years ago) for each rock unit.

11. The igneous rock in rock unit B was formed _____.
12. The igneous rock in rock unit C was formed _____.
13. The metamorphic rock in rock unit G was formed _____.

Step 3: Index fossils have been found in the sedimentary layers! Use **Figure 9.11** to help you to identify the periods in which layers A, D, E, and F were formed as well as their age in mya based on the discovery of the following fossils:

14. The following fossils were found in Layer A:



Figure 9.13

Layer A was formed _____ mya during the _____ Period.

15. The following fossils were found in Layer D:



Figure 9.14

Layer D was formed _____ mya during the _____ Period.

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SA Adler



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Table of Contents Lab Exercise Dating Rock Strata

1. Understanding the eBook Lab Exercise Dating Rock Strata
 - The Rise of Digital Reading Lab Exercise Dating Rock Strata
 - Advantages of eBooks Over Traditional Books
2. Identifying Lab Exercise Dating Rock Strata
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Lab Exercise Dating Rock Strata
 - User-Friendly Interface
4. Exploring eBook Recommendations from Lab Exercise Dating Rock Strata
 - Personalized Recommendations
 - Lab Exercise Dating Rock Strata User Reviews and Ratings
 - Lab Exercise Dating Rock Strata and Bestseller Lists
5. Accessing Lab Exercise Dating Rock Strata Free and Paid eBooks
 - Lab Exercise Dating Rock Strata Public Domain eBooks
 - Lab Exercise Dating Rock Strata eBook Subscription Services
 - Lab Exercise Dating Rock Strata Budget-Friendly Options

6. Navigating Lab Exercise Dating Rock Strata eBook Formats
 - ePub, PDF, MOBI, and More
 - Lab Exercise Dating Rock Strata Compatibility with Devices
 - Lab Exercise Dating Rock Strata Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Lab Exercise Dating Rock Strata
 - Highlighting and Note-Taking Lab Exercise Dating Rock Strata
 - Interactive Elements Lab Exercise Dating Rock Strata
8. Staying Engaged with Lab Exercise Dating Rock Strata
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Lab Exercise Dating Rock Strata
9. Balancing eBooks and Physical Books Lab Exercise Dating Rock Strata
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Lab Exercise Dating Rock Strata
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Lab Exercise Dating Rock Strata
 - Setting Reading Goals Lab Exercise Dating Rock Strata
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Lab Exercise Dating Rock Strata
 - Fact-Checking eBook Content of Lab Exercise Dating Rock Strata
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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