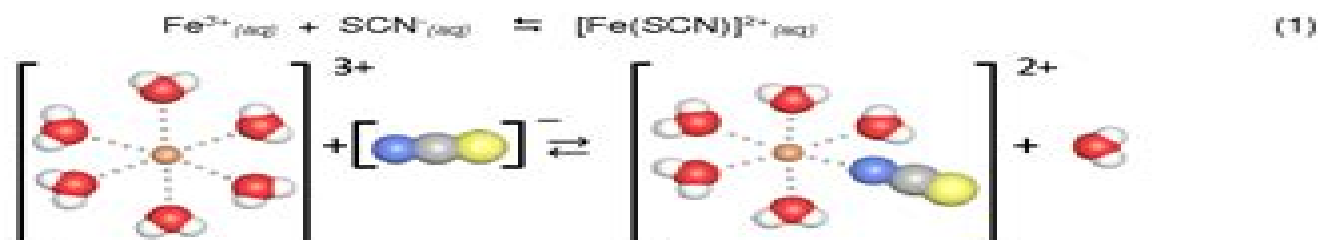


Determining an Equilibrium Constant

Introduction

The equilibrium relationship we are studying in this lab is the aqueous reaction between iron(III) and the thiocyanate ion. The net ionic equation and chemistry are:



The equilibrium constant expression is:

$$K_c = \frac{[\text{Fe}(\text{SCN})^{2+}]}{[\text{Fe}^{3+}][\text{SCN}^{-}]} \quad (2)$$

This experiment uses a Spec-200 spectrophotometer to measure the molar concentration of $[\text{Fe}(\text{SCN})]^{2+}_{(aq)}$, and the instrument needs to be calibrated before data collection. You will use your Spec-200 instrument to measure absorption data for a series of solutions of known $[\text{Fe}(\text{SCN})]^{2+}_{(aq)}$ concentration. This should give you a linear plot of the absorbance with respect to the $[\text{Fe}(\text{SCN})]^{2+}_{(aq)}$ concentration in accord with Beer's Law. An explanation of spectrophotometry and Beer's law is included in this lab as Appendix 1.

The solutions of known $[\text{Fe}(\text{SCN})]^{2+}_{(aq)}$ concentration provided to you are prepared using Le Chatelier's principle—we put a known amount of $\text{SCN}^{-}_{(aq)}$ in solution, and then added a very large excess of $\text{Fe}^{3+}_{(aq)}$ to the solution, forcing the equilibrium to essentially favor quantitative production of $[\text{Fe}(\text{SCN})]^{2+}$.

Procedure

Calibrating the Spec-200 for $[\text{Fe}(\text{SCN})]^{2+}$.

The procedure begins by calibrating the Spec-200 instruments for λ_{max} and the absorbance as a function of $[\text{Fe}(\text{SCN})]^{2+}$ concentration. We have done this procedure before for other labs.

1. Turn on the spectrophotometer and follow the on screen commands. When the spectrophotometer has finished initializing:
 - a. Press the enter key (↵) to select "SPEC 200E Modern Interface"
 - b. With the word "Application" Highlighted press the right arrow (→) three times to select the "Scan" Application
 - c. Press the down arrow (↓) twice to highlight "Low λ ." If the value is not set to 400 adjust it using the left and right arrow keys. If it is set to 400 press the down arrow to highlight "High λ ."
 - d. Use the left and right arrow keys to adjust this value to 700
 - e. Press the down arrow once to highlight "Next" and press enter. The spectrophotometer is now ready for use.
2. The "blank" for zeroing the spec-200 instrument will be DI water. The DI water should fill at least 2/3 of the cuvette with no bubbles. Use a plastic transfer pipet.
3. Insert the blank into the spectrophotometer and press the zero button (0.00). When the instrument finishes zeroing (the screen reads 0.00) remove the cuvette.

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Patrick Vollmar

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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