CHAPTER 18 HOW CHEMICALS MIX

Multiple Choice

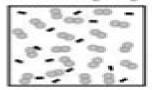
Identify the choice that best completes the statement or answers the question.

- 1. A combination of two or more substances in which they no longer retain their chemical properties is called
 - a. heterogeneous mixture

d. mixture

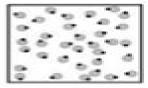
b. compound c. suspension

- e. periodic trend.
- The following image represents which kind of matter?



- a. a mixture
- b. an element
- c. a compound

- d. none of the above.
- e. all of the above
- The following image represents which kind of matter?



- a. an element
- b. a compound
- c. a mixture

- d. none of the above
- e. all of the above

- 4. Which of the following is a mixture?
 - m. sault
 - b. iron

 - c. gold

- d. belium
- e. air
- 5. How does a suspension differ from a solution?
 - A suspension is a heterogeneous mixture consisting of different phases whereas a solution is a homogeneous mixture consisting of a single phase.
 - b. The difference between a suspension and a solution can only be determined by chemical mornages.
 - c. Although a solution and suspension are both homogeneous mixtures, only the components of a suspension will separate by spinning the mixture in a centrifuge.
 - d. A suspension is a heterogeneous mixture whose components can be separated by simple filtration. A solution is a homogeneous mixture which cannot be separated by simple

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