

Name: _____

Honors Chemistry Unit 1 Study Guide: Chapters 4, 11, & 19

1. What particles are found in the nucleus of an atom? What is the charge of the nucleus? What is the charge of the atom? It would be helpful to draw and label this.

In the nucleus, there are protons and neutrons, the charge is positive, a charge of an atom is neutral

2. Draw and label the parts found within a box on the periodic table.

3. How is an atom's atomic number related to its number of protons? Number of electrons?

The atom's number is the same as it's number of protons and electrons

4. What is the charge of the nucleus for element atomic number 56?

It is positive charge

5. Which particles account for most of an atom's mass?

The protons and neutrons

6. How is mass number related to the number of protons and neutrons an atom has?

The mass number is the sum of protons and neutrons

7. When you do not know the units of mass, what can you use? Write the full name and abbreviation.

You can use amu or atomic mass units

8. Distinguish between Democritus' idea and Dalton's atomic theory.

9. What parts of Dalton's atomic theory are now known to be false?

10. Draw and label the five atomic models discussed in class.

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student can use simple straightforward strategies to start making A s in their courses and enjoy a lifetime of deep effective
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Resources in Education, 1998 **General Electric Review** General Electric Company, 1929

Directory of Distance Learning Opportunities Modoc Press, Inc., 2003-02-28 This book provides an overview of current K 12 courses and programs offered in the United States as correspondence study or via such electronic delivery systems as satellite cable or the Internet The Directory includes over 6 000 courses offered by 154 institutions or distance learning consortium members Following an introduction that describes existing practices and delivery methods the Directory offers three indexes Subject Index of Courses Offered by Level Course Level Index Geographic Index All information was supplied by the institutions Entries include current contact information a description of the institution and the courses offered grade level and admission information tuition and fee information enrollment periods delivery information equipment requirements credit and grading information library services and accreditation

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