

ACID-BASE STUDY GUIDE

- Normal Plasma pH 7.35 to 7.45 (hydrogen (H⁺) concentration)
 - Higher this goes, higher the mortality rate
 - Measure the acidity or alkalinity of blood
 - Homeostatic mechanisms keep pH normal
 - Buffer systems, kidneys & lungs
 - Greater H⁺ concentration = lower pH & more acidic solute
 - Lower H⁺ concentration = higher pH & more alkaline solute
- Major ECF buffer system
 - Bicarbonate-carbonic acid system
 - $\text{CO}_2 + \text{H}_2\text{O} \leftrightarrow \text{H}_2\text{CO}_3 \leftrightarrow \text{H}^+ \text{ and } \text{HCO}_3^-$
 - Kidneys regulate bicarbonate in the ECF
 - Lungs (controlled by medulla) regulate CO₂, thus the carbonic acid in the ECF
 - Lungs are first responders
- Other buffer systems
 - ECF: inorganic phosphates and plasma proteins
 - ICF: proteins, organic, inorganic phosphates
 - Hemoglobin

METABOLIC ACIDOSIS

pH lower than 7.35

- Can occur by a gain of H⁺ or loss of bicarbonate ions (<22 mm Hg)
- Causes
 - Direct loss of bicarbonate (diarrhea, administration of IV nutrition w/o bicarb or bicarb-producing solutes, diuretic use, early renal insufficiency, excess chloride, lower intestinal fistulas, ureterostomies)
 - Most common due to kidney injury
 - Lactic acidosis (higher it is, the worse sepsis is), salicylate poisoning, renal failure, methanol, ethylene, or propylene toxicity, DKA, ketoacidosis due to starvation
 - Systemic issue: diabetes
- Clinical Manifestations
 - Headache, confusion, drowsiness, increased RR/depth, low BP, nausea, vomiting, cold/clammy skin, arrhythmias, shock, peripheral vasodilation, decreased cardiac output
- Medical Management
 - First have to correct underlying issue

Key Acid Base Study Guide

JE Gale



Key Acid Base Study Guide:

Unveiling the Power of Verbal Art: An Psychological Sojourn through **Key Acid Base Study Guide**

In a world inundated with monitors and the cacophony of fast communication, the profound energy and mental resonance of verbal art often fade into obscurity, eclipsed by the constant onslaught of noise and distractions. However, located within the lyrical pages of **Key Acid Base Study Guide**, a interesting function of literary splendor that pulses with natural thoughts, lies an wonderful trip waiting to be embarked upon. Written with a virtuoso wordsmith, that enchanting opus guides readers on a psychological odyssey, softly revealing the latent possible and profound affect embedded within the complex internet of language. Within the heart-wrenching expanse of the evocative analysis, we will embark upon an introspective exploration of the book is central subjects, dissect its captivating writing type, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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